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=> file registry
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STRUCTURE FILE UPDATES: 29 MAR 2009 HIGHEST RN 1129300-01-1 DICTIONARY FILE UPDATES: 29 MAR 2009 HIGHEST RN 1129300-01-1

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http://www.cas.org/support/stngen/stndoc/properties.html

1-60 5-62 6-61 7-59 8-58 9-57 10-52 11-68 14-54 15-66

16-67 18-63 19-64

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20-65 22-45 24-71 27-75 28-76 29-77 30-78 31-89 32-90 33-91 34-51 38-48
39-87 40-88
41-81 42-53 43-80 44-79 46-82 47-48 47-49 47-50 51-85 51-86 52-55 52-56
53-83 53-84
54-69 54-70 71-72 71-73 73-74
ring bonds :
1-2 \quad 1-6 \quad 2-3 \quad 2-7 \quad 3-4 \quad 3-10 \quad 4-5 \quad 5-6 \quad 7-8 \quad 8-9 \quad 9-10 \quad 11-12 \quad 11-16 \quad 12-13 \quad 12-17
13 - 14 \quad 13 - 20 \quad 14 - 15 \quad 15 - 16 \quad 17 - 18 \quad 18 - 19 \quad 19 - 20 \quad 21 - 22 \quad 21 - 26 \quad 22 - 23 \quad 23 - 24 \quad 24 - 25
25-26 25-27
26-30 27-28 28-29 29-30 31-32 31-36 32-33 33-34 34-35 35-36 35-37 36-40
37-38 38-39
39-40 41-42 41-46 42-43 43-44 44-45 45-46
exact bonds :
1-60 \quad 5-62 \quad 6-61 \quad 7-59 \quad 8-58 \quad 9-57 \quad 10-52 \quad 11-68 \quad 14-54 \quad 15-66 \quad 16-67 \quad 18-63 \quad 19-64
20-65 \quad 22-45 \quad 24-71 \quad 27-75 \quad 28-76 \quad 29-77 \quad 30-78 \quad 31-89 \quad 32-90 \quad 33-91 \quad 34-51 \quad 38-48
39-87 40-88
41-81 42-53 43-80 44-79 46-82 47-48 47-49 47-50 51-85 51-86 52-55 52-56
53-83 53-84
54-69 54-70 73-74
normalized bonds :
1-2 \quad 1-6 \quad 2-3 \quad 2-7 \quad 3-4 \quad 3-10 \quad 4-5 \quad 5-6 \quad 7-8 \quad 8-9 \quad 9-10 \quad 11-12 \quad 11-16 \quad 12-13 \quad 12-17
13 - 14 \quad 13 - 20 \quad 14 - 15 \quad 15 - 16 \quad 17 - 18 \quad 18 - 19 \quad 19 - 20 \quad 21 - 22 \quad 21 - 26 \quad 22 - 23 \quad 23 - 24 \quad 24 - 25
25-26 25-27
26-30 27-28 28-29 29-30 31-32 31-36 32-33 33-34 34-35 35-36 35-37 36-40
37-38 38-39
39-40 41-42 41-46 42-43 43-44 44-45 45-46 71-72 71-73
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G1:[*1],[*2],[*3],[*4]

Match level:

1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:Atom 8:Atom 9:Atom 10:Atom 11:Atom 12:Atom 13:Atom 14:Atom 15:Atom 16:Atom 17:Atom 18:Atom 19:Atom 20:Atom 21:Atom 22:Atom 23:Atom 24:Atom 25:Atom 26:Atom 27:Atom 28:Atom 29:Atom 30:Atom 31:Atom 32:Atom 33:Atom 34:Atom 35:Atom 36:Atom 37:Atom 38:Atom 39:Atom 40:Atom 41:Atom 42:Atom 43:Atom 44:Atom 45:Atom 46:Atom 47:CLASS 48:CLASS 49:CLASS 50:CLASS 51:CLASS 52:CLASS 53:CLASS 54:CLASS 63:CLASS 64:CLASS 65:CLASS 66:CLASS 67:CLASS 68:CLASS 69:CLASS 70:CLASS 71:CLASS 73:CLASS 74:CLASS 75:CLASS 76:CLASS 77:CLASS 78:CLASS 79:CLASS 80:CLASS 81:CLASS 82:CLASS 83:CLASS 84:CLASS 85:CLASS 86:CLASS 87:CLASS 88:CLASS 90:CLASS 91:CLASS 97:CLASS

=> file zcaplus

FILE 'ZCAPLUS' ENTERED AT 15:46:06 ON 31 MAR 2009
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FILE COVERS 1907 - 31 Mar 2009 VOL 150 ISS 14 FILE LAST UPDATED: 30 Mar 2009 (20090330/ED)

ZCAplus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2008.

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http://www.cas.org/legal/infopolicy.html

This file contains CAS Registry Numbers for easy and accurate substance identification.

'OBI' IS DEFAULT SEARCH FIELD FOR 'ZCAPLUS' FILE

=> d stat	que L	47
L35	18168	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON WANG B?/AU
L36	8823	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON GAO X?/AU
L37	14705	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON YANG W?/AU
L38	3780	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON FANG H?/AU
L39	7303	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON YAN Y?/AU
L40	273	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L35 AND (L36 OR L37
		OR L38 OR L39)
L41	65	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L36 AND (L37 OR L38
		OR L39)
L42	36	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L37 AND (L38 OR L39)
L43	11	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L38 AND L39
L44	15	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L40 AND (L41 OR L42
		OR L43)
L45	1	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L41 AND (L42 OR L43)
L46	0	SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L42 AND L43
L47	15	SEA FILE-ZCAPLUS SPE-ON ABB-ON PLU-ON (L44 OR L45 OR L46)
П д /	10	SEA FILE-ZCAFLOS SFE-ON ADD-ON FLO-ON (L44 OK L45 OK L40)

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=> d stat que L48
L1 STR
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* STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT *

Structure attributes must be viewed using STN Express query preparation.

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L6
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L7
           143 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L6
L35
         18168 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON WANG B?/AU
L36
         8823 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON GAO X?/AU
L37
         14705 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON YANG W?/AU
          3780 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON FANG H?/AU
L38
L39
          7303 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON YAN Y?/AU
L48
             8 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L7 AND (L35 OR L36 OR
               L37 OR L38 OR L39)
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L51 22 L47 OR L48

=> file medline embase biosis wpix FILE 'MEDLINE' ENTERED AT 15:46:48 ON 31 MAR 2009

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=> d stat que L50
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L36
          8823 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON GAO X?/AU
         14705 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON YANG W?/AU
L37
          3780 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON FANG H?/AU 7303 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON YAN Y?/AU
L38
L39
           273 SEA FILE-ZCAPLUS SPE-ON ABB-ON PLU-ON L35 AND (L36 OR L37
L40
                 OR L38 OR L39)
L41
             65 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L36 AND (L37 OR L38
                OR L39)
             36 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L37 AND (L38 OR L39)
L42
             11 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L38 AND L39
15 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L40 AND (L41 OR L42
L43
L44
                OR L43)
             1 SEA FILE-ZCAPLUS SPE-ON ABB-ON PLU-ON L41 AND (L42 OR L43)
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L46
             O SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L42 AND L43
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L47
             19 SEA L47
L50
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=> dup rem L51 L50

FILE 'ZCAPLUS' ENTERED AT 15:47:14 ON 31 MAR 2009

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PROCESSING COMPLETED FOR L51
PROCESSING COMPLETED FOR L50
L52
24 DUP REM L51 L50 (17 DUPLICATES REMOVED)
ANSWERS '1-22' FROM FILE ZCAPLUS
ANSWERS '23-24' FROM FILE BIOSIS

=> d ibib abs hitstr L52 1-22; d iall L52 23-24

L52 ANSWER 1 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 1 ACCESSION NUMBER: 2006:180182 ZCAPLUS $\underline{\text{Full-text}}$

DOCUMENT NUMBER: 144:407472

TITLE: Substituent effect on anthracene-based bisboronic acid

alucose sensors

AUTHOR(S): Kaur, Gurpreet; Fang, Hao; Gao, Mingming; Li,

Haibo; Wang, Binghe

CORPORATE SOURCE: Department of Chemistry and Center for Biotechnology

and Drug Design, Georgia State University, Atlanta,

GA, 30302-4098, USA

SOURCE: Tetrahedron (2006), 62(11), 2583-2589

CODEN: TETRAB; ISSN: 0040-4020

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 144:407472

AB Earlier the authors communicated an anthracene-based bisboronic acid sensor for glucose. Aimed at understanding the substituent effect, the authors have introduced various functional groups, such as the cyano, nitro, and fluoro group on the boronic acid moiety of this glucose sensor. Fluorescent binding studies indicated that the cyano-substituted sensor has the highest affinity (K 2540 M-1) for glucose, but the lowest selectivity (three-fold over fructose); the fluoro-substituted compound shows the lowest affinity (630 M-1) and a modest selectivity (15-fold over fructose); and the unsubstituted one shows the highest selectivity over fructose (43-fold) and a modest affinity (1472 M-1).

REFERENCE COUNT: 51 THERE ARE 51 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 2 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 2

ACCESSION NUMBER: 2005:239227 ZCAPLUS Full-text

DOCUMENT NUMBER: 142:308776

TITLE: Water soluble boronic acid fluorescent reporter

compounds and methods of use thereof

INVENTOR(S): Wang, Binghe; Gao, Xingming; Yang, Wengian;

Fang, Hao; Yan, Jun

PATENT ASSIGNEE(S): USA

SOURCE: PCT Int. Appl., 80 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PAT	CENT 1	NO.			KIN:	D	DATE			APPL	ICAT		DATE						
	WO 2005024416						_	2005	0317	,	WO 2	004-	 US28	 838		20040907				
	W: AE, AG, A		AL,	AM,	ΑT,	AU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,				
			CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,		
			GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KΖ,	LC,		
			LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NI,		
			NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,		
			ΤJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW		
		RW:	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,		
			ΑZ,	BY,	KG,	KΖ,	MD,	RU,	ΤJ,	TM,	AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,		
			EE,	ES,	FΙ,	FR,	GB,	GR,	HU,	IE,	ΙΤ,	LU,	MC,	NL,	PL,	PT,	RO,	SE,		
			SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,		
			SN,	TD,	ΤG															
	US	2007	0274	922		A1		2007	1129		US 2	007-	5708	07		2	0070.	531		
PRIO	RITY	APP:	LN.	INFO	.:						US 2	003-	5007	85P]	P 2	0030	905		
										,	WO 2	004-	US28	838	Ī	W 2	0040	907		
	- ~ -								0000											

OTHER SOURCE(S): MARPAT 142:308776

AB Described herein are boronic acid fluorescent compds. and methods of use thereof.

IT 86-58-8, 8-Quinolineboronic acid 355386-94-6,

5-Quinolinylboronic acid 373384-17-9,

2-(4-Boronophenyl)-4-quinolinecarboxylic acid

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)

RN 373384-17-9 ZCAPLUS

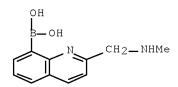
CN 4-Quinolinecarboxylic acid, 2-(4-boronophenyl)- (9CI) (CA INDEX NAME)

IT 847862-01-5P

RL: ARG (Analytical reagent use); PRP (Properties); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

RN 847862-01-5 ZCAPLUS

CN Boronic acid, B-[2-[(methylamino)methyl]-8-quinolinyl]- (CA INDEX NAME)



THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 1 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 3 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 3

ACCESSION NUMBER: 2004:324501 ZCAPLUS Full-text

DOCUMENT NUMBER: 142:130006

TITLE: The First Fluorescent Diboronic Acid Sensor Specific

for Hepatocellular Carcinoma Cells Expressing Sialyl

Lewis X

Yang, Wengian; Fan, Haiying; Gao, Xingming; Gao, AUTHOR(S):

> Shouhai; Karnati, Vishnu Vardhan Reddy; Ni, Weijuan; Hooks, W. Borden; Carson, John; Weston, Brent; Wang,

Binghe

Department of Chemistry, Georgia State University, CORPORATE SOURCE:

Atlanta, GA, 30303, USA

SOURCE: Chemistry & Biology (2004), 11(4), 439-448

CODEN: CBOLE2; ISSN: 1074-5521

PUBLISHER: Cell Press DOCUMENT TYPE: Journal LANGUAGE: English

CASREACT 142:130006 OTHER SOURCE(S):

Carbohydrate antigens with subterminal fucosylation have been implicated in the development and progression of several cancers, including hepatocellular carcinoma (HCC). Fluorescent sensors targeting fucosylated carbohydrate antigens could potentially be used for diagnostic and other applications. The authors have designed and synthesized a series of 26 diboronic acid compds. as potential fluorescent sensors for such carbohydrates. Among these compds., 7q was able to fluorescently label cells expressing high levels of sLex (HEPG2) within a concentration range of 0.5 to 10 μM . This compound (7q) did not label cells expressing Lewis Y (HEP3B), nor cells without fucosylated antigens (COS7). This represents the first example of a fluorescent compound labeling cells based on cell surface carbohydrate structures.

REFERENCE COUNT: THERE ARE 44 CITED REFERENCES AVAILABLE FOR THIS 44RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 4 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 4

ACCESSION NUMBER: 2003:177702 ZCAPLUS Full-text

DOCUMENT NUMBER: 139:16782

TITLE: The first fluorescent sensor for D-glucarate based on

the cooperative action of boronic acid and guanidinium

AUTHOR(S): Yang, Wenqian; Yan, Jun; Fang, Hao; Wang, Binghe CORPORATE SOURCE:

Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

Chemical Communications (Cambridge, United Kingdom) SOURCE:

(2003), (6), 792-793

CODEN: CHCOFS; ISSN: 1359-7345

PUBLISHER: Royal Society of Chemistry

DOCUMENT TYPE: Journal LANGUAGE: English

AB A new fluorescent sensor with a recognition unit consisting of a boronic acid moiety and a guanidinium unit shows selective binding of D-glucarate in

aqueous solution

REFERENCE COUNT: 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 5 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 5

ACCESSION NUMBER: 2003:341171 ZCAPLUS Full-text

DOCUMENT NUMBER: 139:316357

TITLE: Boronic acid compounds as potential pharmaceutical

agents

AUTHOR(S): Yang, Wenqian; Gao, Xingming; Wang, Binghe CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Medicinal Research Reviews (2003), 23(3), 346-368

CODEN: MRREDD; ISSN: 0198-6325

PUBLISHER: John Wiley & Sons, Inc. DOCUMENT TYPE: Journal; General Review

LANGUAGE: English

AB A review. Boronic acid compds. have been used, because of their unique structural features, for the development of potent enzyme inhibitors, boron neutron capture agents for cancer therapy, and as antibody mimics that recognize biol. important saccharides. Consequently, there has been a surge of interests in boronic acid compds. This study reviews the recent

development in this area during the last six years.

REFERENCE COUNT: 122 THERE ARE 122 CITED REFERENCES AVAILABLE FOR

THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

L52 ANSWER 6 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 6

ACCESSION NUMBER: 2002:640931 ZCAPLUS Full-text

DOCUMENT NUMBER: 138:136693

TITLE: Catechol pendant polystyrene for solid-phase synthesis

AUTHOR(S): Yang, Wengian; Gao, Xingming; Springsteen, Greq;

Wang, Binghe

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Tetrahedron Letters (2002), 43(36), 6339-6342

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 138:136693

AB A catechol-pendant polystyrene was prepared from the Merrifield resin via a convenient procedure with high-d. loading. Due to the high affinity binding between catechol and boronic acid, the polymer resin readily captures boronic acid compds. The feasibility of using immobilized catechol to capture boronic acid products for purification and solid-phase transformation was demonstrated. Moreover, the immobilized catechol was also used for the preparation of resin-bound catecholborane, which can be used as a solid-phase amidation reagent.

REFERENCE COUNT: 35 THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 7 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 7

ACCESSION NUMBER: 2002:846231 ZCAPLUS Full-text

DOCUMENT NUMBER: 138:267844

TITLE: A glucose-selective fluorescence sensor based on

boronic acid-diol recognition

AUTHOR(S): Karnati, Vishnu Vardhan; Gao, Xingming; Gao,

Shouhai; Yang, Wengian; Ni, Weijuan; Sankar,

Sabapathy; Wang, Binghe

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Bioorganic & Medicinal Chemistry Letters (2002),

12(23), 3373-3377

CODEN: BMCLE8; ISSN: 0960-894X

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB A glucose selective diphenylboronic acid fluorescent sensor (10a) with a Ka of 1472 M-1 has been synthesized and evaluated. This sensor shows a 43- and 49-fold selectivity for glucose over fructose and galactose, resp. The binding affinity improvement is about 300-fold and the selectivity improvement for glucose over fructose is about 1400-fold compared with the monoboronic acid compound, phenylboronic acid. 1H NMR studies indicate that sensor 10a binds with α -d-glucofuranose in a bidentate manner (1:1 ratio).

REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 8 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN DUPLICATE 8

ACCESSION NUMBER: 2002:543670 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 137:306853

TITLE: Diboronic acids as fluorescent probes for cells

expressing sialyl lewis X

AUTHOR(S): Yang, Wenqian; Gao, Shouhai; Gao, Kingming;

Karnati, Vishnu Vardhan Reddy; Ni, Weijuan; Wang, Binghe; Hooks, W. Borden; Carson, John; Weston, Brent

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Bioorganic & Medicinal Chemistry Letters (2002),

12(16), 2175-2177

CODEN: BMCLE8; ISSN: 0960-894X

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB A series of fluorescent diboronic acids was synthesized in nine steps as potential sensors for sially Lewis X (sLex). The fluorescent binding studies of these sensors with sLex were carried out in a mixed aqueous solution Compound 7e was found to show the strongest fluorescence enhancement upon binding with sLex. Using cell cultures, 7e was shown to label sLex-expressing HEPG2 cells at 1 μ M, while non-sLex-expressing cells were not labeled.

REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 9 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2009:277789 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 150:292251

TITLE: ARGO-YBJ constraints on very high energy emission from

GRBs

AUTHOR(S): Aielli, G.; Bacci, C.; Bartoli, B.; Bernardini, P.;

Bi, X. J.; Bleve, C.; Branchini, P.; Budano, A.; Bussino, S.; Calabrese Melcarne, A. K.; Camarri, P.; Cao, Z.; Cappa, A.; Cardarelli, R.; Catalanotti, S.; Cattaneo, C.; Cavaliere, S.; Celio, P.; Chen, S. Z.; Chen, Y.; Cheng, N.; Creti, P.; Cui, S. W.; Cusumano,

G.; Dai, B. Z.; D'Ali Staiti, G.; Danzengluobu;

Dattoli, M.; De Mitri, I.; D'Ettorre Piazzoli, B.; De

Vincenzi, M.; Di Girolamo, T.; Ding, X. H.; Di Sciascio, G.; Feng, C. F.; Feng, Zhaoyang; Feng,

Zhenyong; Ferrigno, C.; Galeazzi, F.; Galeotti, P.; Gao, X. Y.; Gargana, R.; Gou, Q. B.; Guo, Y. Q.; He, H. H.; Hu, Haibing; Hu, Hongbo; Huang, Q.; Iacovacci, M.; Iuppa, R.; James, I.; Jia, H. Y.; Labaciren; Li, H. J.; Li, J. Y.; Li, X. X.; Li, Y. R.; Liberti, B.; Liguori, G.; Liu, C.; Liu, C. Q.; Liu, M. Y.; Liu, J.; Lu, H.; Ma, X. H.; Mancarella, G.; Mari, S. M.; Marsella, G.; Martello, D.; Mastroianni, S.; Meng, X. R.; Mu, J.; Nicastro, L.; Ning, C. C.; Palummo, L.; Panareo, M.; Perrone, L.; Pistilli, P.; Qu, X. B.; Rossi, E.; Ruggieri, F.; Saggese, L.; Salvini, P.; Santonico, R.; Segreto, A.; Shen, P. R.; Sheng, X. D.; Shi, F.; Stanescu, C.; Surdo, A.; Tan, Y. H.; Vallania, P.; Vernetto, S.; Vigorito, C.; Wang, B.; Wang, H.; Wang, Y. G.; Wu, C. Y.; Wu, H. R.; Xu, B.; Xue, L.; Yan, Y. X.; Yang, H. T.; Yang, Q. Y.; Yang, X. C.; Yu, G. C.; Yuan, A. F.; Zha, M.; Zhang, H. M.; Zhang, J. L.; Zhang, L.; Zhang, P.; Zhang, X. Y.; Zhang, Y.; Zhaxisangzhu; Zhou, X. X.; Zhu, F. R.; Zhu, O. O.; Zizzi, G.

CORPORATE SOURCE: Dipartimento di Fisica, Universita "Tor Vergata",

Rome, 00133, Italy

SOURCE: arXiv.org, e-Print Archive, Astrophysics (2009) 1-13,

arXiv:0903.0119v1 [astro-ph.HE], 1 Mar 2009

CODEN: AARSC7

URL: http://arxiv.org/PS_cache/arxiv/pdf/0903/0903.011

9v1.pdf

PUBLISHER: Cornell University Library

DOCUMENT TYPE: Preprint LANGUAGE: English

The ARGO-YBJ experiment is designed for very high energy γ -astronomy and cosmic ray researches. With a large sensitive area fully covered with resistive plate chambers at a very high altitude (4300 m a.s.l.), the ARGO-YBJ detector is used to search for transient phenomena, such as γ -ray bursts (GRB). Because the ARGO-YBJ detector has a large field of view (FOV .apprx. 2 sr) and is operated in a full duty cycle (>90%), it is one of the best ground-based GRB surveying apparatuses. Working at a relatively high energy threshold around few hundred GeV, the ARGO-YBJ detector is operated in searches for high energy GRBs following alarms set by satellite borne observations at lower energies. A sensitivity of the ARGO-YBJ detector for GRB detection is estimated Upper limits of fluence with 99% confidence level of 31 GRBs inside the FOV from June 2006 to Jan. 2009 are set in two energy ranges of 10 GeV-100 GeV and 10 GeV-1 TeV.

L52 ANSWER 10 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2009:264669 ZCAPLUS Full-text

TITLE: Compositions for regulating or modulating quorum sensing in bacteria, methods of using the compounds,

and methods of regulating or modulating quorum sensing

in bacteria

INVENTOR(S): Wang, Bingha; Ni, Nanting; Wang, Junfeng; Lu,

Chung-Dar; Chou, Han-Ting; Li, Minyong; Zheng,

Shilong; Cheng, Yunfeng; Peng, Hanjing

PATENT ASSIGNEE(S): Georgia State University Research Foundation, Inc.,

IISA

SOURCE: PCT Int. Appl., 75pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATEN:	r NO.			KIN	D	DATE			APPL	ICAT	ION I		DATE			
WO 20	A2	_	 2009	0305	,	WO 2	008-		20080606							
W	: AE,	AG,	AL,	AM,	AO,	ΑT,	ΑU,	ΑZ,	BA,	BB,	BG,	BH,	BR,	BW,	BY,	BZ,
	CA,	CH,	CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DO,	DZ,	EC,	EE,	EG,	ES,
	FI,	GB,	GD,	GE,	GH,	GM,	GT,	HN,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,
	KG,	ΚM,	KN,	KP,	KR,	KΖ,	LA,	LC,	LK,	LR,	LS,	LT,	LU,	LY,	MA,	MD,
	ME,	MG,	MK,	MN,	MW,	MX,	MY,	MZ,	NA,	NG,	NΙ,	NO,	NΖ,	OM,	PG,	PH,
	PL,	PT,	RO,	RS,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SM,	SV,	SY,	ТJ,	TM,
	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	ZA,	ZM,	ZW			
RI	W: AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FΙ,	FR,	GB,	GR,	HR,	HU,
	IE,	IS,	ΙΤ,	LT,	LU,	LV,	MC,	MT,	NL,	NO,	PL,	PT,	RO,	SE,	SI,	SK,
	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	${ m ML}$,	MR,	NE,	SN,	TD,
	TG,	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,
	AM,	ΑZ,	BY,	KG,	KΖ,	MD,	RU,	ТJ,	MT							

PRIORITY APPLN. INFO.:

US 2007-933735P P 20070608

The present disclosure encompasses compds. and compns. that are useful as specific AI-2 antagonists for the control of bacterial quorum sensing. Although the AI-2 antagonists according to the present disclosure may not have bactericidal effect, their ability to attenuate virulence, drug resistance, and/or biofilm formation have therapeutic benefits. In addition, the AI-2 antagonists of the present disclosure can also be used as tools to probe bacterial AI-2 functions. The present disclosure also encompasses methods for inhibiting or attenuating microbial virulence, biofilm formation, and drug resistance. The methods are suitable for preventing bacteria from accruing and forming extensive biofilms that may be a health or hygiene hazard or a phys. issue, such as in the blockage of water or fuel lines.

IT 355386-94-6

RL: BSU (Biological study, unclassified); BIOL (Biological study) (compns. for regulating or modulating quorum sensing in bacteria)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)

L52 ANSWER 11 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2009:87071 ZCAPLUS Full-text

DOCUMENT NUMBER: 150:163129

TITLE: Nucleotides and aptamers containing boronic acid

groups having biased binding to glycosylated proteins,

and uses for glycoprotein detection

INVENTOR(S): Wang, Binghe; Li, Minyong

PATENT ASSIGNEE(S): Georgia State University Research Foundation, Inc.,

USA

SOURCE: PCT Int. Appl., 94pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PAT:	ENT	NO.			KIN	D	DATE	DATE APPLICATION NO.								DATE			
WO .	 2009	 0123	 63		A2	_	 2009	 0122	•	 WO 2	 008-	 US70	 288		20080717				
	W:	ΑE,	AG,	AL,	AM,	AO,	ΑT,	ΑU,	ΑZ,	BA,	BB,	ВG,	BH,	BR,	BW,	BY,	BZ,		
		CA,	CH,	CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DO,	DZ,	EC,	EE,	EG,	ES,		
		FI,	GB,	GD,	GE,	GH,	GM,	GT,	HN,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	ΚE,		
		KG,	KM,	KN,	KP,	KR,	KΖ,	LA,	LC,	LK,	LR,	LS,	LT,	LU,	LY,	MA,	MD,		
		ME,	MG,	MK,	MN,	MW,	MX,	MY,	MZ,	NA,	NG,	NI,	NO,	NZ,	OM,	PG,	PH,		
		PL,	PT,	RO,	RS,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SM,	ST,	SV,	SY,	ТJ,		
		TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UΖ,	VC,	VN,	ZA,	ZM,	ZW				
	RW:	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	HR,	HU,		
		ΙE,	IS,	IT,	LT,	LU,	LV,	MC,	MT,	NL,	NO,	PL,	PT,	RO,	SE,	SI,	SK,		
		TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,	MR,	NE,	SN,	TD,		
		ΤG,	BW,	GH,	GM,	ΚE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,		
		ΑM,	ΑZ,	BY,	KG,	KΖ,	MD,	RU,	ТJ,	TM									
DTTV	ΛDD	T M	TNEO							110 2	007_	9506	Q 1 D		D 2	0070	710		

PRIORITY APPLN. INFO.:

US 2007-950681P P 20070719

OTHER SOURCE(S): MARPAT 150:163129

AB The present disclosure encompasses oligonucleotide aptamers selectively binding a target glycosylated polypeptide or protein, and having biased affinity for the glycan through a boronic acid linked to a nucleosidic base of a nucleotide(s). The disclosure further encompasses methods for isolating an aptamer(s) selectively binding a target glycosylated polypeptide, where, from a population of randomized oligonucleotides that have at least one nucleotide having a boronic acid label linked to a base, is selected a first subpopulation of aptamers binding to the target glycosylated polypeptide or protein. This subpopulation is then amplified without using boronic acid—modified TTP, and amplification products not binding to a target glycosylated polypeptide or protein are selected. The second subpopulation of aptamers is then amplified using boronic acid—modified TTP to provide a population of boronic acid—modified aptamers capable of selectively binding to a glycosylation site of a target polypeptide or protein. Other aspects of the

IT 86-58-8 355386-94-6

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (fluorescent boronic acid compound; nucleotides and aptamers containing boronic acid groups having biased binding to glycosylated proteins, and uses for glycoprotein detection)

disclosure encompass methods for the use of the modified aptamers to detect

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

glycosylated species of a polypeptide or protein.

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)



SOURCE:

PUBLISHER:

AUTHOR(S):

L52 ANSWER 12 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2008:1401248 ZCAPLUS Full-text

TITLE: Analysis of polycyclic aromatic hydrocarbons in

environmental air

AUTHOR(S): Nie, Jian-Qun; Wang, Bao-Xing; Yang, Wei-Zu; Yan,

Yang; Ying, Hou; Chen, Guo-Hui; Xu, Wang

CORPORATE SOURCE: Yunnan Reascend Tobacco Technology (Group) Co., Ltd.,

Kunming, 650106, Peop. Rep. China Yancao Keji (2008), (10), 25-29

CODEN: YKAEA8; ISSN: 1002-0861

Zhongguo Yancao Keji Xinxi Zhongxin

DOCUMENT TYPE: Journal LANGUAGE: Chinese

In order to investigate the concentration and constitution of polycyclic AΒ aromatic hydrocarbons (PAHs) in environmental air, the sampling method for 16 PAHs in air was optimized, and the PAHs in the air sampled from 5representative public places (including panel testing room, teahouse, restaurant, cross-roads and Karaoke compartment) in Kunming with the method were analyzed by GC/MS under selected ion monitoring (SIM) mode. The results showed that: 1) it was preferable to collect PAHs in air with silica gel adsorbent for 120 min at the air flow rate of 1.0 L/min; 2) except for those from panel testing room, dibenzo[a,h]anthrene was not detectable in all samples; 3) the places in the order of total content of PAHs in air were panel testing room > teahouse > restaurant > crossroads > Karaoke compartment; 4) except for teahouse, the constitution and distribution of PAHs in environmental air in all tested places were similar, while the concentration of each PAH was different; 5) the contents of acenaphthene, fluoranthene and pyrene in the air of teahouse were obviously higher than those in the other 4 places.

L52 ANSWER 13 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2007:863170 ZCAPLUS Full-text

DOCUMENT NUMBER: 147:469384

TITLE: A unique quinolineboronic acid-based supramolecular

structure that relies on double intermolecular B-N

bonds for self-assembly in solid state and in

solution. [Erratum to document cited in CA146:501102] Zhang, Yanling; Li, Minyong; Chandrasekaran, Sekar;

Gao, Xingming; Fang, Xikui; Lee, Hsiau-Wei; Hardcastle, Kenneth; Yang, Jenny; Wang, Binghe

CORPORATE SOURCE: Department of Chemistry and Center for Biotechnology

and Drug Design, Georgia State University, Atlanta,

GA, 30302-4089, USA

SOURCE: Tetrahedron (2007), 63(37), 9256

CODEN: TETRAB; ISSN: 0040-4020

PUBLISHER: Elsevier Ltd.

DOCUMENT TYPE: Journal

LANGUAGE: English

The mol. radii presented were not normalized for concentration effect though the radii at different concns. for the internal reference, dioxane, were presented side by side for comparison. Presenting the normalized radii will allow readers to see a more accurate picture of the size effect. Using dioxane, which has a hydrodynamic radius of 2.12Å, as a size reference, the effective hydrodynamic radii for all samples have been normalized. Supplemental Table 2 summarizes the results of mol. radii (Å) for 8-QBA and 5-QBA after internal reference normalization. It is clear that the mol. radius of 8-QBA is 20% greater than that of 5-QBA at all concns. These results are consistent with self-association of 8-QBA into dimer (methanol), however, with no obvious concentration dependence observed

IT 86-58-8, 8-Quinolineboronic acid 355386-94-6,

5-Quinolineboronic acid RL: PRP (Properties)

(crystal structure; unique quinolineboronic acid-based supramol. structure that relies on double intermol. boron-nitrogen bonds for self-assembly in solid state and in solution (Erratum))

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)

L52 ANSWER 14 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2007:301956 ZCAPLUS Full-text

DOCUMENT NUMBER: 146:501102

TITLE: A unique quinolineboronic acid-based supramolecular

structure that relies on double intermolecular B-N bonds for self-assembly in solid state and in solution

AUTHOR(S): Zhang, Yanling; Li, Minyong; Chandrasekaran, Sekar;

Gao, Xingming; Fang, Xikui; Lee, Hsiau-Wei; Hardcastle, Kenneth; Yang, Jenny; Wang, Binghe

CORPORATE SOURCE: Department of Chemistry and Center for Biotechnology

and Drug Design, Georgia State University, Atlanta,

GA, 30302-4089, USA

SOURCE: Tetrahedron (2007), 63(16), 3287-3292

CODEN: TETRAB; ISSN: 0040-4020

PUBLISHER: Elsevier Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

The boronic acid functional group plays very important roles in sugar recognition, catalysis, organic synthesis, and supramol. assembly. Therefore, understanding the unique properties of this functional group is very important. 8-Quinolineboronic acid (8-QBA) is capable of self-assembling in solid state through a unique intermol. B-N bond mechanism reinforced by intermol. boronic anhydride formation, π - π stacking, and hydrogen bond formation. NMR NOE and diffusion studies indicate that intermol. B-N interaction also exists in solution with 8-QBA. In contrast, a positional isomer of 8-QBA, 5-quinolineboronic acid (5-QBA) showed very different behaviors in crystal packing and in solution and therefore different supramol. network. Understanding the structural features of this unique 8-QBA assembly could be very helpful for the future design of new sugar sensors, mol. catalysts, and supramol. assemblies.

IT 86-58-8, 8-Quinolineboronic acid 355386-94-6

RL: PRP (Properties)

(crystal structure; unique quinolineboronic acid-based supramol. structure that relies on double intermol. boron-nitrogen bonds for self-assembly in solid state and in solution)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)

REFERENCE COUNT: 66 THERE ARE 66 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 15 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2006:1245011 ZCAPLUS Full-text

DOCUMENT NUMBER: 147:449674

TITLE: Design and manufacturing of a composite lattice

structure reinforced by continuous carbon fibers

AUTHOR(S): Fan, Hualin; Yang, Wei; Wang, Bin; Yan, Yong;

Fu, Qiang; Fang, Daining; Zhuang, Zhuo

CORPORATE SOURCE: Department of Engineering Mechanics, Tsinghua

University, Beijing, 100084, Peop. Rep. China

SOURCE: Tsinghua Science and Technology (2006), 11(5), 515-522

CODEN: TSTEF7; ISSN: 1007-0214

PUBLISHER: Tsinghua University Press

DOCUMENT TYPE: Journal LANGUAGE: English

AB New techniques have been developed to make materials with a periodic three-dimensional lattice structure. The high stiffness per unit weight and multifunction of such lattice structures make them attractive for use in aeronautic and astronautic structures. In this paper, epoxy-soaked continuous carbon fibers were first introduced to make lattice composite structures, which maximize the specific load carrying capacity. A micromech. anal. of several designs, each corresponding to a different manufacturing route, was carried out, in order to find the optimized lattice structure with maximum specific stiffness. An intertwining method was chosen and developed as the best route to make lattice composite materials reinforced by carbon fibers. A sandwich-weaved sample with a three-dimensional intertwined lattice structure core was found to be best. The manufacturing of such a composite lattice material was outlined. In addition to a high shear strength of the core and the integral manufacturing method, the lattice sandwich structure is expected to possess better mech. capability.

REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 16 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2006:358117 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 146:72834

TITLE: Design and experiment of high-current low-pressure

plasma-cathode e-qun

AUTHOR(S): Xie, Wen-kai; Li, Xiao-yun; Wang, Bin; Meng, Lin;

Yan, Yang; Gao, Xin-yan

CORPORATE SOURCE: Institute of High Energy Electronics, University of

Electronics Science and Technology of China, Chengdu,

610054, Peop. Rep. China

SOURCE: Qiangjiguang Yu Lizishu (2006), 18(2), 235-240

CODEN: QYLIEL; ISSN: 1001-4322 Qiangjiquang Yu Lizishu Bianjibu

DOCUMENT TYPE: Journal LANGUAGE: Chinese

PUBLISHER:

AB The preliminary design of a new high-power low-pressure plasma-cathode e-gun is presented. Based on the hollow cathode effect and low-pressure glow discharge empirical formulas, the hollow cathode, the accelerating gap, and the working gas pressure region are given. The general exptl. device of the low-pressure plasma cathode electron-gun generating high c.d. e-beam source is shown. Expts. was done in continuous filled-in gases and gases-puff condition, and the discharging current of 150-200~A, the width of $60~\mu\text{s}$ and the collector current of 30-80~A, the width of $60~\mu\text{s}$ are obtained. The new plasma cathode e-gun can take the place of material cathode e-gun, especially in plasma filled microwave tubes.

L52 ANSWER 17 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:1119491 ZCAPLUS Full-text

DOCUMENT NUMBER: 144:18975

TITLE: A new type of boronic acid fluorescent reporter

compound for sugar recognition

AUTHOR(S): Yang, Wenqian; Lin, Li; Wang, Binghe

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Tetrahedron Letters (2005), 46(46), 7981-7984

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

Fluorescent boronic acids that change fluorescent properties upon carbohydrate

binding are very useful for the preparation of fluorescent sensors for sugars.

Herein the authors report 5-quinolineboronic acid (5-QBA) that shows significant fluorescent property changes through a unique pKa-switching

mechanism upon binding a diol in aqueous solution

ΙT 355386-94-6

RL: ARG (Analytical reagent use); BUU (Biological use, unclassified); ANST

(Analytical study); BIOL (Biological study); USES (Uses)

(new type of boronic acid fluorescent reporter compound for sugar

recognition)

355386-94-6 ZCAPLUS RN

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)



THERE ARE 54 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 54

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 18 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN 2007:266839 ZCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 147:427389

TITLE: Biological and medicinal applications of boronic acids

Yang, Wenqian; Gao, Xingming; Wang, Binghe AUTHOR(S):

CORPORATE SOURCE: JRIX Pharmaceuticals, Inc., Florence, SC, 29501, USA

Boronic Acids (2005), 481-512. Editor(s): Hall, SOURCE:

Dennis G. Wiley-VCH Verlag GmbH & Co. KGaA: Weinheim,

Germany.

CODEN: 69IZIO; ISBN: 978-3-527-30991-7

DOCUMENT TYPE: Conference; General Review

LANGUAGE: English

A review.

REFERENCE COUNT: 186 THERE ARE 186 CITED REFERENCES AVAILABLE FOR

THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

L52 ANSWER 19 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN 2005:6485 ZCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 142:459422

TITLE: A new type of water-soluble fluorescent boronic acid

suitable for construction of polyboronic acids for

carbohydrate recognition

Yang, Wenqian; Lin, Li; Wang, Binghe AUTHOR(S):

Department of Chemistry, North Carolina State CORPORATE SOURCE:

University, Raleigh, NC, 27695-8204, USA

SOURCE: Heterocyclic Communications (2004), 10(6), 383-388

PUBLISHER:

CODEN: HCOMEX; ISSN: 0793-0283 Freund Publishing House Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB In this paper the authors report boronic acid 1 with a quinoline moiety as a new type of fluorescent probe for carbohydrates, which shows significant fluorescence intensity increases upon sugar binding at physiol. pH. This compound has the unique structural feature of separating the boronic acid moiety from the presumed fluorophore, and is ready for the construction of polyboronic acids through tethering to its carboxylic group for high selectivity and affinity recognition of carbohydrates of biol. interest.

II 373384-17-9

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (water-soluble fluorescent boronic acid with quinoline moiety for carbohydrate recognition)

RN 373384-17-9 ZCAPLUS

CN 4-Quinolinecarboxylic acid, 2-(4-boronophenyl)- (9CI) (CA INDEX NAME)

REFERENCE COUNT: 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 20 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2003:215761 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 139:130191

TITLE: A novel type of fluorescent boronic acid that shows

large fluorescence intensity changes upon binding with a carbohydrate in aqueous solution at physiological pH Yang, Wangian; Yan, Jun; Springsteen, Greg; Deeter,

AUTHOR(S): Yang, Wenqian; Yan, Susan; Wang, Binghe

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Bioorganic & Medicinal Chemistry Letters (2003),

13(6), 1019-1022

CODEN: BMCLE8; ISSN: 0960-894X

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

AB In this paper we report 8-quinoline boronic acid as a novel type of fluorescent probe for carbohydrates. This boronic acid responds to the binding of a carbohydrate with over 40-fold increases in fluorescence intensity and shows optimal fluorescence change at physiol. pH in aqueous solution

IT 86-58-8, 8-Quinolineboronic acid 86-58-8D,

8-Quinolineboronic acid, esters

RL: ARG (Analytical reagent use); PRP (Properties); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent); USES (Uses)

(fluorescent 8-quinoline boronic acid that shows large fluorescence

intensity changes upon binding with a carbohydrate in aqueous solution at physiol. pH)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

86-58-8 ZCAPLUS RN

Boronic acid, B-8-quinolinyl- (CA INDEX NAME) CN



THERE ARE 40 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 40 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L52 ANSWER 21 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN 2002:190790 ZCAPLUS Full-text ACCESSION NUMBER:

A diboronic acid fluorescence sensor with high TITLE:

affinity and selectivity for glucose

Wang, Singhe; Gao, Xingming; Karnati, Vishnu AUTHOR(S):

Vardhan Reddy; Yang, Wengian; Sankar, Sabapathy; Ni,

Weijuan

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Abstracts of Papers, 223rd ACS National Meeting,

Orlando, FL, United States, April 7-11, 2002 (2002), ORGN-255. American Chemical Society: Washington, D.

С.

CODEN: 69CKQP

DOCUMENT TYPE: Conference; Meeting Abstract

LANGUAGE: English

Fifty diboronic acid-based fluorescent sensors have been synthesized. One (5a) of them showed 43- and 49-fold selectivity for glucose over fructose and galactose, resp. Examination of the complex between 5-a and D-glucose by NMR and ESI-MS indicates the formation of a 1:1 complex in the glucofuranose form.

L52 ANSWER 22 OF 24 ZCAPLUS COPYRIGHT 2009 ACS on STN 2002:618012 ZCAPLUS Full-text ACCESSION NUMBER:

Fluorescent tags for cells expressing sialyl Lewis X TITLE: AUTHOR(S):

Wang, Binghe; Yang, Wenqian; Gao, Shouhai; Gao,

Xingming; Karnati, Vishnu Vardhan Reddy; Ni, Weijuan;

Hooks, W. Borden; Carson, John; Weston, Brent

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Abstracts of Papers, 224th ACS National Meeting,

> Boston, MA, United States, August 18-22, 2002 (2002), MEDI-160. American Chemical Society: Washington, D.

CODEN: 69CZPZ

DOCUMENT TYPE: Conference; Meeting Abstract

English LANGUAGE:

It is well known that the expression and over-expression of certain cell surface carbohydrates are associated with the development of some cancer. Aimed at developing compds. that recognize certain cell-surface carbohydrates with high specificity and affinity, we have designed and synthesized a series of fluorescent anthracene diboronic acids as potential sensors for sialyl Lewis X (sLex). One of these compds. showed a strong fluorescence enhancement upon binding with sLex. Further biol. studies showed that this sensor compound was able to label sLex-expressing HEPG2 cells at 1 micromolar, while the non-expressing control cells were not stained.

L52 ANSWER 23 OF 24 BIOSIS COPYRIGHT (c) 2009 The Thomson Corporation on

STN

2002:510926 BIOSIS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: PREV200200510926

Fluorescent tags for cells expressing sialyl Lewis X. TITLE: Wang, Binghe [Reprint author]; Yang, Wengian [Reprint AUTHOR(S): author]; Gao, Shouhai [Reprint author]; Gao, Xingming [Reprint author]; Karnati, Vishnu Vardhan Reddy [Reprint author]; Ni, Weijuan [Reprint author]; Hooks, W. Borden;

Carson, John; Weston, Brent

CORPORATE SOURCE: Department of Chemistry, North Carolina State University,

Raleigh, NC, 27695-8204, USA

binghe_wang@ncsu.edu

SOURCE: Abstracts of Papers American Chemical Society, (2002) Vol.

224, No. 1-2, pp. MEDI 160. print.

Meeting Info.: 224th National Meeting of the American Chemical Society. Boston, MA, USA. August 18-22, 2002.

CODEN: ACSRAL. ISSN: 0065-7727.

DOCUMENT TYPE: Conference; (Meeting)

Conference; Abstract; (Meeting Abstract)

LANGUAGE: English

ENTRY DATE: Entered STN: 2 Oct 2002

Last Updated on STN: 2 Oct 2002

CONCEPT CODE: General biology - Symposia, transactions and proceedings

00520

Cytology - General 02502 02508 Cytology - Human

Biochemistry studies - General

Pathology - Therapy 12512 Pharmacology - General 22002

Pharmacology - Clinical pharmacology

Neoplasms - Pathology, clinical aspects and systemic

effects 24004

INDEX TERMS: Major Concepts

Biochemistry and Molecular Biophysics; Cell Biology;

Methods and Techniques; Oncology (Human Medicine,

Medical Sciences); Pharmacology

INDEX TERMS: Diseases

cancer: neoplastic disease, drug therapy

Neoplasms (MeSH)

INDEX TERMS: Chemicals & Biochemicals

Sialyl Lewis X: expression; fluorescent anthracene

diboronic acids: cellular fluorescent tag,

pharmacodynamics, sensor, sialyl Lewis X detector,

synthesis

INDEX TERMS: Methods & Equipment

chemical synthesis: Synthetic Techniques, pharmacological method, synthetic method

INDEX TERMS: Miscellaneous Descriptors

drug development; Meeting Abstract

ORGANISM: Classifier

Hominidae 86215

Super Taxa

Primates; Mammalia; Vertebrata; Chordata; Animalia

Organism Name HEPG2 cell line

Taxa Notes

Animals, Chordates, Humans, Mammals, Primates,

Vertebrates

REGISTRY NUMBER: 98603-84-0 (Sialyl Lewis X)

 ${\tt L52}$ $\,$ ANSWER 24 OF 24 $\,$ BIOSIS COPYRIGHT (c) 2009 The Thomson Corporation on

STN

ACCESSION NUMBER: 2002:529933 BIOSIS Full-text

DOCUMENT NUMBER: PREV200200529933

TITLE: A diboronic acid fluorescence sensor with high affinity and

selectivity for glucose.

AUTHOR(S): Wang, Binghe [Reprint author]; Gao, Xingming [Reprint

author]; Vardhan, Vishnu [Reprint author]; Karnati, Reddy
[Reprint author]; Yang, Wanquin [Reprint author]; Sankar,
Sabapathy [Reprint author]; Ni, Weijun [Reprint author]

CORPORATE SOURCE: Department of Chemistry, North Carolina State University,

Raleigh, NC, 27695-8204, USA

binghe_wang@ncsu.edu

SOURCE: Abstracts of Papers American Chemical Society, (2002) Vol.

223, No. 1-2, pp. ORGN 255. print.

Meeting Info.: 223rd National Meeting of the American Chemical Society. Orlando, FL, USA. April 07-11, 2002.

CODEN: ACSRAL. ISSN: 0065-7727.

DOCUMENT TYPE: Conference; (Meeting)

Conference; Abstract; (Meeting Abstract)

LANGUAGE: English

ENTRY DATE: Entered STN: 16 Oct 2002

Last Updated on STN: 16 Oct 2002

CONCEPT CODE: General biology - Symposia, transactions and proceedings

00520

Biochemistry studies - General 10060

Biochemistry studies - Carbohydrates 10068

INDEX TERMS: Major Concepts

Biochemistry and Molecular Biophysics

INDEX TERMS: Chemicals & Biochemicals

D-glucose; diboronic acid; diboronic acid fluorescence

sensor; fructose; galactose

INDEX TERMS: Miscellaneous Descriptors

molecular interaction; Meeting Abstract

REGISTRY NUMBER: 50-99-7 (D-glucose)

13675-18-8 (diboronic acid)

57-48-7Q (fructose) 30237-26-4Q (fructose) 59-23-4Q (galactose) 26566-61-0Q (galactose)

chain bonds :

=> file registry
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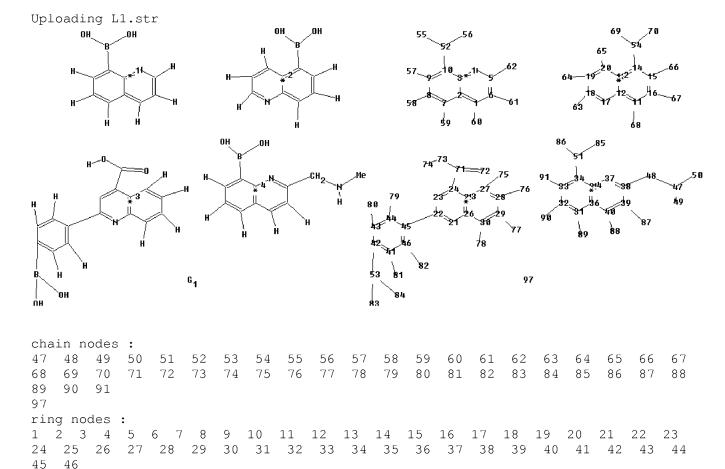
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http://www.cas.org/support/stngen/stndoc/properties.html



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1-60 \quad 5-62 \quad 6-61 \quad 7-59 \quad 8-58 \quad 9-57 \quad 10-52 \quad 11-68 \quad 14-54 \quad 15-66 \quad 16-67 \quad 18-63 \quad 19-64
20-65 22-45 24-71 27-75 28-76 29-77 30-78 31-89 32-90 33-91 34-51 38-48
39-87 40-88
41-81 42-53 43-80 44-79 46-82 47-48 47-49 47-50 51-85 51-86 52-55 52-56
53-83 53-84
54-69 54-70 71-72 71-73 73-74
ring bonds :
1-2^{-1} 1-6 2-3 2-7 3-4 3-10 4-5 5-6 7-8 8-9 9-10 11-12 11-16 12-13 12-17
13 - 14 \quad 13 - 20 \quad 14 - 15 \quad 15 - 16 \quad 17 - 18 \quad 18 - 19 \quad 19 - 20 \quad 21 - 22 \quad 21 - 26 \quad 22 - 23 \quad 23 - 24 \quad 24 - 25
25-26 25-27
26-30 27-28 28-29 29-30 31-32 31-36 32-33 33-34 34-35 35-36 35-37 36-40
37-38 38-39
39-40 41-42 41-46 42-43 43-44 44-45 45-46
exact bonds :
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20-65 \quad 22-45 \quad 24-71 \quad 27-75 \quad 28-76 \quad 29-77 \quad 30-78 \quad 31-89 \quad 32-90 \quad 33-91 \quad 34-51 \quad 38-48 \quad 39-79 \quad 39-7
39-87 40-88
41-81 42-53 43-80 44-79 46-82 47-48 47-49 47-50 51-85 51-86 52-55 52-56
53-83 53-84
54-69 54-70 73-74
normalized bonds :
1-2 \quad 1-6 \quad 2-3 \quad 2-7 \quad 3-4 \quad 3-10 \quad 4-5 \quad 5-6 \quad 7-8 \quad 8-9 \quad 9-10 \quad 11-12 \quad 11-16 \quad 12-13 \quad 12-17
13 - 14 \quad 13 - 20 \quad 14 - 15 \quad 15 - 16 \quad 17 - 18 \quad 18 - 19 \quad 19 - 20 \quad 21 - 22 \quad 21 - 26 \quad 22 - 23 \quad 23 - 24 \quad 24 - 25
25-26 25-27
26-30 27-28 28-29 29-30 31-32 31-36 32-33 33-34 34-35 35-36 35-37 36-40
37-38 38-39
39-40 41-42 41-46 42-43 43-44 44-45 45-46 71-72 71-73
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G1:[*1],[*2],[*3],[*4]

Match level :

1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:Atom 7:Atom 8:Atom 9:Atom 10:Atom 11:Atom 12:Atom 13:Atom 14:Atom 15:Atom 16:Atom 17:Atom 18:Atom 19:Atom 20:Atom 21:Atom 22:Atom 23:Atom 24:Atom 25:Atom 26:Atom 27:Atom 28:Atom 29:Atom 30:Atom 31:Atom 32:Atom 33:Atom 34:Atom 35:Atom 36:Atom 37:Atom 38:Atom 39:Atom 40:Atom 41:Atom 42:Atom 43:Atom 44:Atom 45:Atom 46:Atom 47:CLASS 48:CLASS 49:CLASS 50:CLASS 51:CLASS 52:CLASS 53:CLASS 54:CLASS 55:CLASS 56:CLASS 57:CLASS 58:CLASS 59:CLASS 60:CLASS 61:CLASS 62:CLASS 63:CLASS 64:CLASS 65:CLASS 66:CLASS 67:CLASS 68:CLASS 69:CLASS 70:CLASS 71:CLASS 73:CLASS 74:CLASS 75:CLASS 76:CLASS 77:CLASS 78:CLASS 79:CLASS 80:CLASS 81:CLASS 82:CLASS 83:CLASS 84:CLASS 85:CLASS 86:CLASS 87:CLASS 88:CLASS 90:CLASS 91:CLASS 97:CLASS

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FILE COVERS 1907 - 31 Mar 2009 VOL 150 ISS 14 FILE LAST UPDATED: 30 Mar 2009 (20090330/ED)

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=> d stat que L12 L1STR

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Structure attributes must be viewed using STN Express query preparation.

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3 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON L6 NOT (L9 OR L10) L11

L12 5 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L11

=> d ibib abs hitstr L12 1-5

L12 ANSWER 1 OF 5 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2008:928331 ZCAPLUS Full-text

DOCUMENT NUMBER: 149:288705

TITLE: Solution Phase Synthesis of a Diverse Library of

Highly Substituted Isoxazoles

Waldo, Jesse P.; Mehta, Saurabh; Neuenswander, AUTHOR(S):

> Benjamin; Lushington, Gerald H.; Larock, Richard C. Department of Chemistry, Iowa State University, Ames,

CORPORATE SOURCE:

IA, 50011, USA

SOURCE: Journal of Combinatorial Chemistry (2008), 10(5),

658-663

CODEN: JCCHFF; ISSN: 1520-4766

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

CASREACT 149:288705 OTHER SOURCE(S):

The iodocyclization of O-methyloximes of 2-alkyn-1-ones afforded 4iodoisoxazoles, which underwent various palladium-catalyzed reactions to yield 3,4,5-trisubstituted isoxazoles. The palladium-catalyzed processes have been adapted to parallel synthesis utilizing com. available boronic acid,

acetylene, styrene, and amine sublibraries. Accordingly, a diverse 51-member library of 3,4,5-trisubstituted isoxazoles has been generated.

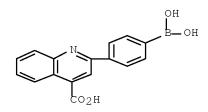
IT 373384-17-9

RL: RCT (Reactant); RACT (Reactant or reagent)

(solution phase synthesis of a diverse library of highly substituted isoxazoles)

RN 373384-17-9 ZCAPLUS

CN 4-Quinolinecarboxylic acid, 2-(4-boronophenyl)- (9CI) (CA INDEX NAME)



REFERENCE COUNT: 50 THERE ARE 50 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L12 ANSWER 2 OF 5 ZCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2005:239227 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 142:308776

TITLE: Water soluble boronic acid fluorescent reporter

compounds and methods of use thereof

INVENTOR(S): Wang, Binghe; Gao, Xingming; Yang, Wenqian; Fang, Hao;

Yan, Jun

PATENT ASSIGNEE(S): USA

SOURCE: PCT Int. Appl., 80 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA	TENT I	NO.			KIN	D	DATE			APPL	ICAT	ION I		DATE				
WO	2005	0244	16		A1 20050317			1	WO 2	004-	JS28	20040907						
	W: AE, AG, AL,				AM,	ΑT,	ΑU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,	
		CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,	
		GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KΖ,	LC,	
		LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NI,	
		NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,	
		ТJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW	
	RW:	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,	
		AΖ,	BY,	KG,	KΖ,	MD,	RU,	ΤJ,	TM,	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	
		EE,	ES,	FI,	FR,	GB,	GR,	HU,	ΙE,	ΙΤ,	LU,	MC,	NL,	PL,	PT,	RO,	SE,	
		SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,	
		SN,	TD,	TG														
US	2007	0274	922		A1		2007	1129	1	US 2	007-	5708	07	20070531				
PRIORIT	Y APP					1	US 2003-500785P			85P	P 20030905							
						1	WO 2	004-	JS28	838	W 20040907							
OTHER S	OURCE	(S):			MAR	PAT	142:	3087	76									
3.0	1.3	3 3	,					1 1 6	- 7				1			1	_	

AB Described herein are boronic acid fluorescent compds. and methods of use thereof.

IT 373384-17-9, 2-(4-Boronophenyl)-4-quinolinecarboxylic acid RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)

(analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

RN 373384-17-9 ZCAPLUS

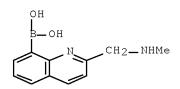
CN 4-Quinolinecarboxylic acid, 2-(4-boronophenyl)- (9CI) (CA INDEX NAME)

IT 847862-01-5P

RL: ARG (Analytical reagent use); PRP (Properties); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

RN 847862-01-5 ZCAPLUS

CN Boronic acid, B-[2-[(methylamino)methyl]-8-quinolinyl]- (CA INDEX NAME)



REFERENCE COUNT: 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L12 ANSWER 3 OF 5 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:6485 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 142:459422

TITLE: A new type of water-soluble fluorescent boronic acid

suitable for construction of polyboronic acids for

carbohydrate recognition

AUTHOR(S): Yang, Wenqian; Lin, Li; Wang, Binghe

CORPORATE SOURCE: Department of Chemistry, North Carolina State

University, Raleigh, NC, 27695-8204, USA

SOURCE: Heterocyclic Communications (2004), 10(6), 383-388

CODEN: HCOMEX; ISSN: 0793-0283

PUBLISHER: Freund Publishing House Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB In this paper the authors report boronic acid 1 with a quinoline moiety as a new type of fluorescent probe for carbohydrates, which shows significant fluorescence intensity increases upon sugar binding at physiol. pH. This compound has the unique structural feature of separating the boronic acid moiety from the presumed fluorophore, and is ready for the construction of

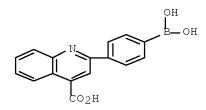
polyboronic acids through tethering to its carboxylic group for high selectivity and affinity recognition of carbohydrates of biol. interest.

IT 373384-17-9

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (water-soluble fluorescent boronic acid with quinoline moiety for carbohydrate recognition)

RN 373384-17-9 ZCAPLUS

CN 4-Quinolinecarboxylic acid, 2-(4-boronophenyl)- (9CI) (CA INDEX NAME)



REFERENCE COUNT: 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L12 ANSWER 4 OF 5 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1959:77801 ZCAPLUS Full-text

DOCUMENT NUMBER: 53:77801
ORIGINAL REFERENCE NO.: 53:14104h-i

TITLE: Acylation of 1-substituted 3,4-dihydroisoquinolines

AUTHOR(S): Gardent, Jean

SOURCE: Compt. rend. (1958), 247, 2010-13

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

1-Substituted 3,4-dihydroisoquinolines are acylated with or without ring-opening, depending on the substituent and the conditions. The cleaved acyl derivative reacts with NH2OH in various ways [Brossi, et al., Chimia 12, 114(1958)]. BzCl-NaOH at 100° with the appropriate 1-substituted-3,4-dihydroisoquinoline yields the following benzoylamines by ring cleavage: 2-benzoylphenylethyl, m. 107°; 2-benzoyl-4,5-diethoxyphenylethyl, m. 127° (oxime, m. 149°); 2-(3',4'-dimethoxyphenylacetyl)-4,5-dimethoxyphenylethyl, m. 133-4°; 2-hexanoyl-4,5-diethoxyphenylethyl, m. 124°. Benzoyl dihydropapaveraldine (open form) m. 168°; N-benzoyldihydropapaverine (closed form) m. 222°; (open form) m. 145-6°; N-benzoyl-1-styryl-6,7-diethoxydihydroisoquinoline (open form), m. 154° (hydroxylamine derivative m. 130°).

IT 127544-62-1

(Derived from data in the 6th Collective Formula Index (1957-1961))

RN 127544-62-1 ZCAPLUS

CN 8-Quinolineboronic acid, hydriodide (6CI) (CA INDEX NAME)

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L12 ANSWER 5 OF 5 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1959:77800 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 53:77800
ORIGINAL REFERENCE NO.: 53:14104d-h

TITLE: Fluoro derivatives of polycyclic carcinogenic

compounds

AUTHOR(S): Bergmann, Ernst D.; Blum, Jochanan; Butanaro, Sara;

Heller, Adam

CORPORATE SOURCE: Hebrew Univ., Jerusalem, Israel SOURCE: Tetrahedron Letters (1959), 1, 15-18

CODEN: TELEAY; ISSN: 0040-4039

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

On the assumption that the determining step in the carcinogenic process in an aromatic hydrocarbon is an electrophilic reaction of the "K" region it is predicted that if the electrophilic substitution takes place only at the fluorinated C atom the carcinogenic activity will decrease but that if substitution takes place only at the nonfluorinated atom or simultaneously on both the fluorinated and nonfluorinated atoms the carcinogenic activity will increase. To test the validity of these theoretical considerations various fluoro derivs. of carcinogenic polycyclics were synthesized. Condensation of 4-FC10H6NH2 with p- and m-MeC6H4NH2 and the N-(p- and -m-tolyl) derivs. (m. 74-5°, b2 187-9°, and m. 53°, b10 208°, resp.) cyclized with ZnCl2 and Ac20 gave 5,7-dimethyl-3-fluoro-1,2-benzacridine (I), m. 174° [picrate, m. 218-20° (decomposition)], and 5.8-dimethyl-3-fluoro-1,2-benzacridine (II), m. $173-5^{\circ}$; picrate, m. 210-13 $^{\circ}$ (decomposition). Treatment of o-C6H4(CO)20 with 4-FC10H6MgBr (from 4-FC10H6Br, m. 36°, b0.8 108°) yielded 53% o-(4-FC10H6CO)C6H4CO2H (III), m. 161° (PhMe). Reaction of III with MeMgBr gave the lactone, m. 140° (alc.), reduced by Zn and HCl or by P and HI in 40 and 100% vields, resp., to o-(4-FC10H6CHMe)C6H4CO2H, m. 170.5-71° (MeNO2), cyclized with H2SO4 to yield 88% 7,12-dihydro-5-fluoro-12-methylbenz[a]anthracen-7-one (III), m. 130°, converted by reduction with Zn dust and aqueous NaOH and subsequent dehydration with alc. HCl to 87% 5-fluoro-12methylbenz[a]anthracene (V), m. 54° (alc.). Treatment of IV with MeMgI gave 76% 5-fluoro-7,12-dimethylbenz[a]anthracene (VI), m. 94° (C6H12). Ultraviolet spectra in CHCl3 are tabulated (compound and λ in m μ (log ϵ) given): I, 274, 283, 285, 296, 342, 369, 390 (4.65, 4.72, 4.75, 4.70, 3.88, 3.92, 3.90); II, 264, 272, 284, 294, 322, 336, 355, 374, 395 (4.52, 4.58, 4.71, 4.64, 3.83, 3.89, 3.83, 3.90, 3.95); V, 285, 295, 348, 396 (4.78, 4.38, 3.79, 3.04); VI, 289, 300, 366 (4.65, 4.68, 3.78).

IT 127544-62-1

(Derived from data in the 6th Collective Formula Index (1957-1961))

RN 127544-62-1 ZCAPLUS

CN 8-Quinolineboronic acid, hydriodide (6CI) (CA INDEX NAME)

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http://www.cas.org/legal/infopolicy.html

This file contains CAS Registry Numbers for easy and accurate substance identification.

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L13
           106 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L9
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L24
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L25
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L26
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L27
L28
L29
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1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 86-58-8 106 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L9

L13

=> d ibib abs hitind hitstr L29 1-4; d ibib abs hitstr L13 102-106

L29 ANSWER 1 OF 4 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:239227 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 142:308776

Water soluble boronic acid fluorescent reporter TITLE:

compounds and methods of use thereof

INVENTOR(S): Wang, Binghe; Gao, Xingming; Yang, Wenqian; Fang, Hao;

Yan, Jun

USA PATENT ASSIGNEE(S):

PCT Int. Appl., 80 pp. SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE: Patent English LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PAT:	ENT 1	. OV			KIN	D	DATE APPLICATION NO.							DATE					
	WO 2005024416					A1	_	20050317		WO 2004-US28838					20040907 <					
		W:	ΑE,	AG,	AL,	AM,	AT,	AU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,		
			CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,		
			GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KΖ,	LC,		
			LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NI,		
			NO,	NΖ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,		
			ΤJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW		
		RW:	BW,	GH,	GM,	ΚE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,		
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			SI,	SK,	TR,	BF,	ΒJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	${ m ML}$,	MR,	NE,		
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PRIO	RITY	APP:	LN.	INFO	.:					US 2003-500785P						P 20030905 <				
										1	WO 2004-US28838					W 20040907				

OTHER SOURCE(S): MARPAT 142:308776

AB Described herein are boronic acid fluorescent compds. and methods of use thereof.

IC ICM G01N033-00

80-3 (Organic Analytical Chemistry) Section cross-reference(s): 9, 33, 64

ST water soluble boronic acid fluorescent indicator

IT Bacterium (genus)

Biosensors

Blood analysis

Cell

```
Fluorescent indicators
     Formation constant
     Lithiation
    Microtiter plates
    Parasite
     Pharmaceutical analysis
     Tumor markers
     Virus
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
    Antibodies and Immunoglobulins
ΙT
     Blood-group substances
     Carbohydrates, analysis
     DNA
     Glycolipids
     Haptens
     Ligands
     Oligosaccharides, analysis
     RNA
     RL: ANT (Analyte); ANST (Analytical study)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
    Nucleic acids
     Oligonucleotides
     Peptides, analysis
     RL: ANT (Analyte); ARG (Analytical reagent use); ANST (Analytical study);
     USES (Uses)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
     Dendritic polymers
     Glycoproteins
     Lipids, uses
     Lipopolysaccharides
     Macromolecular compounds
     Polymers, uses
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
    Acids, uses
ΙT
     Group IIIA element compounds
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (boronic acids; analyte detection by fluorometry with water soluble
        boronic acid fluorescent reporter compds.)
TΤ
     RL: ANT (Analyte); ANST (Analytical study)
        (membrane; analyte detection by fluorometry with water soluble boronic
        acid fluorescent reporter compds.)
     50-70-4, Sorbitol, analysis 50-99-7, D-Glucose, analysis 57-48-7,
ΙT
     D-Fructose, analysis 59-23-4, D-Galactose, analysis 63-42-3, Lactose
     147-81-9, Arabinose 2438-80-4, L-Fucose 3458-28-4, D-Mannose
     17598-81-1, Tagatose 71208-06-5, Lewis X
                                                 82993-43-9 92448-22-1,
     Sialyl Lewis a 98603-84-0, Sialyl Lewis X
     RL: ANT (Analyte); ANST (Analytical study)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
     86-58-8, 8-Quinolineboronic acid
                                      98437-23-1,
ΙT
     2-Benzothienylboronic acid 355386-94-6, 5-Quinolinylboronic acid
     371764-64-6, 4-Quinolinylboronic acid 373384-17-9,
     2-(4-Boronophenyl)-4-quinolinecarboxylic acid 590417-32-6,
     [6-(Dimethylamino)-2-naphthalenyl]-boronic acid
```

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

ΙT 636987-06-9P, 4-(Dimethylamino)naphthaleneboronic acid 847862-01-5P RL: ARG (Analytical reagent use); PRP (Properties); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

4766-33-0P, 5-Amino-1-bromonaphthalene ΙT 5328-76-7P, 5-Nitro-1-bromonaphthalene 10586-45-5P, 5-(Dimethylamino)-1-bromonaphthalene 61047-43-6P, 8-Bromo-2-methylquinoline 847861-96-5P 847861-97-6P

847861-98-7P

847861-99-8P 847862-00-4P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

86-57-7, 1-Nitronaphthalene 121-43-7, Trimethylborate 615-36-1, ΤТ 2-Bromoaniline 4170-30-3, Crotonaldehyde 24424-99-5 59557-93-6, 1-Bromo-4-(dimethylamino)naphthalene 201733-56-4 RL: RCT (Reactant); RACT (Reactant or reagent)

(analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

86-58-8, 8-Quinolineboronic acid ΙT

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

86-58-8 ZCAPLUS RN

Boronic acid, B-8-quinolinyl- (CA INDEX NAME) CN



1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 2 OF 4 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:216822 ZCAPLUS Full-text

DOCUMENT NUMBER: 142:298124

TITLE: Preparation of pyrimidothiophenes as HSP90 inhibitors

INVENTOR(S): Dymock, Brian William; Drysdale, Martin James;

Fromont, Christofe; Jordan, Allan

Vernalis Cambridge Ltd., UK; Cancer Research PATENT ASSIGNEE(S):

Technology Ltd.; The Institute of Cancer Research;

Barril-Alonso, Xavier

PCT Int. Appl., 132 pp. CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

SOURCE:

PATENT NO. KIND DATE APPLICATION NO. DATE

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    WO 2005021552
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                              20050310
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            GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
            LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI,
            NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
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            EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE,
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    CA 2537135
                        Α1
                                                                20040826 <--
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                       A1
PRIORITY APPLN. INFO.:
                                          GB 2003-20300
                                                            A 20030829 <--
                                          GB 2003-27924
                                                            A 20031202 <--
                                          GB 2004-14467
                                                           A 20040629
                                          WO 2004-GB3641
                                                           W 20040826
OTHER SOURCE(S):
                      CASREACT 142:298124; MARPAT 142:298124
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GΙ

Pyrimidothiophenes I [R2 = (Ar1)m-(Alk1)p-Zr-(Alk2)s-Q; Ar1 = (un)substituted aryl, heteroaryl; Alk1, Alk2 = (un)substituted alkylene, alkenylene; m-s = 0, 1; Z = 0, S, COP, CS, SO2, CO2, (un)substituted CONH, CSNH, SO2NH, NHCO, NHSO2, NH; Q = H, carbocyclic, heterocyclic; R3 = H, (un)substituted alkyl, aryl, heteroaryl; R4 = carboxylic ester, carboxamide, sulfonamide] were prepared for use as HSP90 inhibitors. Thus, 2-amino-4,6-dichloro-5-formylpyrimidine was treated with HSCH2CO2Et, followed by Suzuki reaction with PhB(OH)2 to give I [R2 = Ph, R3 = H, R4 = CO2Et] which showed activity in the fluorescence polarization assay for HSP90 inhibition.

IC ICM C07D495-04

ICS A61K031-00; A61P035-00

CC 28-16 (Heterocyclic Compounds (More Than One Hetero Atom))
 Section cross-reference(s): 1

IT 51-45-6, 1H-Imidazole-5-ethanamine 64-04-0, Benzeneethanamine

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86-58-8 91-00-9 98-80-6, Phenylboronic acid 100-51-6, Benzyl
    alcohol, reactions 100-52-7, Benzaldehyde, reactions 104-47-2
    104-78-9 104-88-1, 4-Chlorobenzaldehyde, reactions 105-36-2, Ethyl
    bromoacetate 105-56-6, Ethyl cyanoacetate 106-94-5, 1-Bromopropane
    108-00-9 109-01-3 120-57-0, 1,3-Benzodioxole-5-carboxaldehyde
    138-39-6 140-29-4, Benzeneacetonitrile 351-54-2 585-32-0 623-51-8,
    Ethyl mercaptoacetate 627-35-0 687-51-4 2032-35-1, Bromoacetaldehyde
    diethylacetal 2038-03-1, 4-Morpholineethanamine 2620-50-0,
    1,3-Benzodioxole-5-methanamine 2740-83-2 3731-52-0,
    3-Pyridinemethanamine 4363-35-3, 2-Phenylvinylboronic acid 4439-02-5,
    1,3-Benzodioxole-5-acetonitrile 4510-08-1 4985-46-0 5292-43-3
    5332-73-0 5392-81-4, 2-Diethylaminoethyl bromide 5527-95-7
    5604-46-6, 2-Amino-4,6-dichloro-5-formylpyrimidine 5779-72-6 5980-97-2
    6165-69-1 6238-14-8, 1-Azabicyclo[2.2.2]octan-3-amine 6967-12-0,
    1H-Indazol-6-amine 7663-77-6 14003-16-8 14900-39-1 20173-24-4,
    3-Pyridineethanamine 20845-34-5, (1-Methylpiperidin-2-yl)methanol
    21987-29-1 27757-85-3, 2-Thiophenemethanamine 28739-42-6 29668-44-8
    30433-91-1, 2-Thiopheneethanamine 36276-24-1,
    3-Bromo-4-methylbenzaldehyde 39489-77-5, 2,4-Dichloro-5-nitrophenol
    40299-87-4 53929-74-1 54035-94-8 55499-44-0,
    2,4-Dimethylbenzeneboronic acid 59239-44-0 63503-60-6 73183-34-3
    81731-43-3 86595-37-1 87199-16-4 87199-17-5 94614-83-2
    98437-24-2 100224-74-6, Guanidine carbonate 100379-00-8 113893-08-6
    126401-89-6 126747-14-6 128796-39-4 129271-98-3,
                                                            148839-33-2
    1-Phenylsulfonyl-1H-indole-3-boronic acid 130870-00-7
    153624-46-5 192182-56-2 209919-30-2 214839-25-5 313546-18-8 342408-78-0 351422-73-6 352525-91-8 372963-49-0 385370-80-9 423176-38-9 677743-50-9 847560-47-8 847560-49-0,
    4-Benzyloxy-2-Methylphenylboronic acid 847560-56-9 871231-32-2
    876189-18-3
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of pyrimidothiophenes as HSP90 inhibitors)
    86-58-8
ΙΤ
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of pyrimidothiophenes as HSP90 inhibitors)
    86-58-8 ZCAPLUS
RN
CN
    Boronic acid, B-8-quinolinyl- (CA INDEX NAME)
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REFERENCE COUNT: THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 3 OF 4 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2003:215761 ZCAPLUS Full-text DOCUMENT NUMBER: 139:130191

TITLE: A novel type of fluorescent boronic acid that shows large fluorescence intensity changes upon binding

with a carbohydrate in aqueous solution at

physiological pH

AUTHOR(S): Yang, Wenqian; Yan, Jun; Springsteen, Greg; Deeter,

Susan; Wang, Binghe

10/570807 CORPORATE SOURCE: Department of Chemistry, North Carolina State University, Raleigh, NC, 27695-8204, USA SOURCE: Bioorganic & Medicinal Chemistry Letters (2003), 13(6), 1019-1022 CODEN: BMCLE8; ISSN: 0960-894X PUBLISHER: Elsevier Science B.V. DOCUMENT TYPE: Journal LANGUAGE: English In this paper we report 8-quinoline boronic acid as a novel type of AΒ fluorescent probe for carbohydrates. This boronic acid responds to the binding of a carbohydrate with over 40-fold increases in fluorescence intensity and shows optimal fluorescence change at physiol. pH in aqueous solution 9-5 (Biochemical Methods) CC Section cross-reference(s): 22, 33 ST quinoline boronate fluorescence carbohydrate physiol pH soln ΙT Fluorescence Fluorescent indicators Fluorometry На (fluorescent 8-quinoline boronic acid that shows large fluorescence intensity changes upon binding with a carbohydrate in aqueous solution at physiol. pH) ΙT Carbohydrates, analysis RL: ANT (Analyte); PRP (Properties); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent) (fluorescent 8-quinoline boronic acid that shows large Eluorescence intensity changes upon binding with a carbohydrate in aqueous solution at physiol. pH) 50-99-7, D-Glucose, analysis 57-48-7, D-Fructose, analysis 59-23-4, D-Galactose, analysis 87-81-0, D-Tagatose 5328-37-0, L-Arabinose RL: ANT (Analyte); PRP (Properties); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent) (fluorescent 8-quinoline boronic acid that shows large fluorescence intensity changes upon binding with a carbohydrate in aqueous solution at physiol. pH) 86-58-8, 8-Quinolineboronic acid 86-58-8D, 8-Quinolineboronic acid, esters RL: ARG (Analytical reagent use); PRP (Properties); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent); USES (Uses) (fluorescent 8-quinoline boronic acid that shows large fluorescence intensity changes upon binding with a carbohydrate in aqueous solution at physiol. pH) 86-58-8, 8-Ouinolineboronic acid 86-58-8D, 8-Quinolineboronic acid, esters RL: ARG (Analytical reagent use); PRP (Properties); RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent); USES (Uses) (fluorescent 8-quinoline boronic acid that shows large £luorescence intensity changes upon binding with a carbohydrate in aqueous solution at physiol. pH)

CN

86-58-8 ZCAPLUS

Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

OH N

REFERENCE COUNT: 40 THERE ARE 40 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 4 OF 4 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1970:493281 ZCAPLUS Full-text

DOCUMENT NUMBER: 73:93281

ORIGINAL REFERENCE NO.: 73:15217a,15220a

TITLE: Environmental effects upon the photoluminescence of

8-quinolineboronic acid

AUTHOR(S): Goldman, Michael; Wehry, E. L.

CORPORATE SOURCE: Dep. of Chem., Indiana Univ., Bloomington, IN, USA

SOURCE: Analytical Chemistry (1970), 42(11), 1186-8

CODEN: ANCHAM; ISSN: 0003-2700

DOCUMENT TYPE: Journal LANGUAGE: English

AB The luminescence of 8-quinolineboronic acid (I) has been compared with that of quinoline, as well as that exhibited by 5- and 8-hydroxyquinoline. In contrast to the hydroxyquinolines, singlet \rightarrow triplet intersystem crossing is an important process for I in both hydrocarbon and hydroxylic solvents, implying that a low-lying (n, π^*) excited singlet state plays an important role in determining its luminescence characteristics. The fluorescence efficiency of I is not significantly affected by H-bond-accepting species, but singlet-to-ground internal conversion, presumably involving solvent interactions with the -B(OH)2 group, is an important process for this mol. in hydroxylic media. The luminescence behavior of I is shown to be consistent with the predictions of MO calcns. previously reported.

CC 73 (Spectra by Absorption, Emission, Reflection, or Magnetic Resonance, and Other Optical Properties)

IT Fluorescence

Luminescence

(of quinolineboronic acid, solvent effects on)

IT 86-58-8

RL: PRP (Properties)

(luminescence of, solvent effects on)

IT 86-58-8

RL: PRP (Properties)

(luminescence of, solvent effects on)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)



L13 ANSWER 102 OF 106 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1964:60992 ZCAPLUS Full-text

DOCUMENT NUMBER: 60:60992
ORIGINAL REFERENCE NO.: 60:10705a-c

TITLE: Areneboronic acids with neighboring amine groups

AUTHOR(S): Letsinger, Robert L.

CORPORATE SOURCE: Northwestern Univ., Evanston, IL

SOURCE: Advances in Chemistry Series (1964), 42, 1-16

CODEN: ADCSAJ; ISSN: 0065-2393

DOCUMENT TYPE: Journal LANGUAGE: Unavailable GI For diagram(s), see printed CA Issue.

8-Quinolineboronic acid (I), 2-(2-boronophenyl)benzimidazole (II), and 2-(2-boronobenzyl)benzimidazole were found to catalyze a reaction of chloroethanol with water or alcs. in the presence of collidine. 2-(2-Pyridyl-2-ethynyl)benzeneboronic acid (III), in which the borono and amine functional groups are farther separated than in these compds., did not exhibit comparable activity; however, it underwent isomerization in chloroethanol to a substance which was an active catalyst. The unusual reactivity of these B-N compds. is attributed to cooperation of the boronic acid and amine groups in acting on the hydroxylic substrates. Spectral data relevant to the structures of the compds. are presented and mechanistic pathways for the catalytic reactions are discussed.

IT 86-58-8P, 8-Quinolineboronic acid

RL: PREP (Preparation)
 (preparation of)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)



L13 ANSWER 103 OF 106 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1963:435678 ZCAPLUS Full-text

DOCUMENT NUMBER: 59:35678
ORIGINAL REFERENCE NO.: 59:6426a-b

TITLE: Stereochemistry of the reaction of 8-quinolineboronic

acid with chloro alcohols

AUTHOR(S): Letsinger, Robert L.; Morrison, James D.

CORPORATE SOURCE: Northwestern Univ., Evanston, IL

SOURCE: Journal of the American Chemical Society (1963),

85(15), 2227-9

CODEN: JACSAT; ISSN: 0002-7863

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

The reaction of I with two pairs of stereoisomeric chlorohydrins was investigated. In two-step reactions, involving treatment of the chlorohydrin with I and hydrolysis of the cyclic esters thereby produced, trans-2-chloro-1-indanol was converted to cis-1,2-indandiol, and erythro-2-chloro-1,2-diphenylethanol was converted to dl-hydrobenzoin in good yield. cis-2-Chloro-1-indanol did not undergo carbon-chlorine fission when treated with I under the conditions used for reaction of the trans isomer, and threo-2-chloro-1,2-diphenylethanol afforded in a very slow reaction a low yield of meso-hydrobenzoin. The mechanistic implications of the stereoselectivity of the reaction of 8-quinolineboronic acid are discussed.

IT 86-58-8, 8-Quinolineboronic acid

(reaction with chloro alcs., stereochemistry of)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)



L13 ANSWER 104 OF 106 ZCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1963:435677 ZCAPLUS Full-text

DOCUMENT NUMBER: 59:35677
ORIGINAL REFERENCE NO.: 59:6425h,6426a

TITLE: Organoboron compounds. XIV. Polyfunctional catalysis

by 8-quinolineboronic acid

AUTHOR(S): Letsinger, R. L.; Dandegaonker, S.; Vullo, W. J.;

Morrison, J. D.

CORPORATE SOURCE: Northwestern Univ., Evanston, IL

SOURCE: Journal of the American Chemical Society (1963),

85(15), 2223-7

CODEN: JACSAT; ISSN: 0002-7863

DOCUMENT TYPE: Journal Unavailable

GI For diagram(s), see printed CA Issue.

AB 8-Quinolineboronic acid (I) was found to be a polyfunctional catalyst for hydrolysis of chloroethanol and 3-chloro-1-propanol in dimethylformamide solns. containing water and collidine. In the absence of I, the chloro alcs. underwent slow solvolysis in dimethylformamide solution to products that were not glycols. Both water and ethylene glycol inhibited the catalytic reaction when present in high concentration It is proposed that the boronic acid group in I functions as a binding site for the chloro alc. and that the nitrogen participates in the reaction as a basic or nucleophilic transforming site.

IT 86-58-8, 8-Quinolineboronic acid

(as catalyst in hydrolysis of 2-chloroethanol or 3-chloro-1-propanol)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)



(reaction with chloro alcs., stereochemistry of

L13 ANSWER 105 OF 106 ZCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1960:62736 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 54:62736 ORIGINAL REFERENCE NO.: 54:12140b-c

TITLE: Acid induced rearrangement of 8-isopropyl-1-naphthoic

acid and 8-isopropyl-5,6,7,8-tetrahydro-1-naphthoic acid. Investigation into the nature of the reaction of

8-quinoline-boronic acid and chloro alcohols

AUTHOR(S): Vullo, William J.

CORPORATE SOURCE: Northwestern Univ., Evanston, IL

SOURCE: (1960) 136 pp. Avail.: Univ. Microfilms (Ann Arbor,

Mich.), Order No. 60-460

From: Dissertation Abstr. 20, 3521-2

DOCUMENT TYPE: Dissertation LANGUAGE: Unavailable

AB Unavailable

IT 86-58-8, 8-Quinolineboronic acid (reaction with chloro alcs.)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)



L13 ANSWER 106 OF 106 ZCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1959:77802 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 53:77802

ORIGINAL REFERENCE NO.: 53:14104i,14105a-f

TITLE: Organoboron compounds. IX. 8-Quinolineboronic acid,

its preparation and influence on reactions of

chlorohydrins

AUTHOR(S): Letsinger, Robert L.; Dandegaonker, S. H.

CORPORATE SOURCE: Northwestern Univ., Evanston, IL

SOURCE: Journal of the American Chemical Society (1959), 81,

498-501

CODEN: JACSAT; ISSN: 0002-7863

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

cf. C.A. 53, 9198d. 8-Bromoguinoline (18 g.) in 60 cc. Et20 added at -39° AΒ during 20 min. to 0.216 mole BuLi (cooled to -78°), the mixture stirred 0.5 hr. at -78° , treated during 20 min. dropwise with 76 g. (BuO)3B in 100 cc. Et20 at -39° , stirred 2 hrs., warmed to room temperature overnight, and treated with cold dilute HCl, the aqueous layer neutralized with NaHCO3, and the precipitate recrystd. (EtOH) yielded 11.85 g. 8-quinolineboronic acid (I), m. above 300°. I (0.65 g.) in 15 cc. AcOH and 7 cc. H2O treated at room temperature with 15 cc. 30% H2O2 at room temperature, diluted with H2O, and neutralized with NaHCO3 yielded 0.415 g. 8-hydroxyquinoline, m. 69-70°. I (0.630 g.) heated 36 hrs. with 0.5 g. H2O at 225 $^{\circ}$ in a sealed tube and the product isolated with Et2O yielded quinoline (Ia), identified as the methiodide, m. $133-4^{\circ}$; the aqueous filtrate evaporated gave 0.164 g. B(OH)3. I (0.358 g.) and 0.23 g. o-C6H4(NH2)2 refluxed in 50 cc. C6H6 and evaporated gave 0.50 g. of the corresponding dihydrobenzoboradiazole, m. $188-9^{\circ}$ (CCl4). I with HI gave 92% I.MeI, m. $166-8^{\circ}$. MeI (13.0 g.), 2.70 g. I, and 200 cc. absolute EtOH refluxed 15 hrs. and evaporated gave 92% I.MeI, m. $168-9^{\circ}$ (EtOH). Ia (3.25 g.), 14.85 g. MeI, and 50 cc. EtOH refluxed 18 hrs. and evaporated yielded 6.78 g. methiodide, m. 133°. BuOH (10 g.), 2.9 g. I, and 60 cc. C6H6 distilled azeotropically and worked up gave 2.8 g. Bu ester (II) of I, b4 180° , n25D 1.4840. MeI (5 g.) and 4.93 g. II heated 12 hrs. on the steam bath and evaporated, and the residue washed with Et2O and pentane yielded 7.10 g. II.MeI, hygroscopic solid, m. $70-5^{\circ}$; a 2.21-g. portion steam distilled and the residue evaporated gave 1.81 g. Ia methiodide, m. $135-6^{\circ}$. PhMe (40 cc.), 5 g. Cl(CH2)2OH, b. 127°, n20D 1.4415, and 0.82 g. I partially distilled and cooled gave 1.31 g. ClCH2CH2 ester of I, m. $193-4^{\circ}$ (PhMe). Cl(CH2)2OH and I (8.00 millimoles) each diluted with HCONMe2 to 50.0 cc., the mixture heated at 90.0° , and 5.00 cc. aliquots taken and titrated for chloride ions showed 44.3% reaction after 24 hrs. Cl(CH2)20H without and with added 8.00 millimoles Ia, and with 8.00 millimoles Ia and PbB(OH)2 showed under similar conditions 4.0, 4.3, and 3.4% reaction, resp. Cl(CH2)3OH, b13 62°, n20D 1.4470, and I gave 44.5% reaction, while without or with added 8.00millimoles Ia, the reaction proceeded only to 2.9 and 3.6%, resp. AmCl without and with Ia and with 8.00 millimoles I and BuOH added gave 3.6, 2.2, and 3.5% reaction, resp. The 1st-order rate constant for the reaction of Cl(CH2)4OH, b16 85°, n2OD 1.4520, in HCONMe2 was k 0.09 hr.-1; it is of the order of 60-80 times greater than the rate consts. for Cl(CH2)20H and Cl(CH2)30H. The rate of the chloride ion formation from Cl(CH2)40H (8.00 millimoles) in 50 cc. HCONMe2 at 89.3° with Ia, with Ia and PhB(OH)2, with PhB(OH)2, and with I is represented graphically.

IT %6~58~%, 8-Quinolineboronic acid (and derivs.)

RN 86-58-8 ZCAPLUS

CN Boronic acid, B-8-quinolinyl- (CA INDEX NAME)

=> file registry
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TSCA INFORMATION NOW CURRENT THROUGH January 9, 2009.

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http://www.cas.org/support/stngen/stndoc/properties.html

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FILE COVERS 1907 - 31 Mar 2009 VOL 150 ISS 14 FILE LAST UPDATED: 30 Mar 2009 (20090330/ED)

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http://www.cas.org/legal/infopolicy.html

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'OBI' IS DEFAULT SEARCH FIELD FOR 'ZCAPLUS' FILE

=> d stat que L34

=> d stat que L14

L10 1 SEA FILE=REGISTRY SPE=ON ABB=ON PLU=ON 355386-94-6

L14 49 SEA FILE=ZCAPLUS SPE=ON ABB=ON PLU=ON L10

=> d ibib abs hitind hitstr L34 1; d ibib abs hitstr L14 45-49

L34 ANSWER 1 OF 1 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:239227 ZCAPLUS $\underline{\text{Full-text}}$

DOCUMENT NUMBER: 142:308776

TITLE: Water soluble boronic acid fluorescent reporter

compounds and methods of use thereof

INVENTOR(S): Wang, Binghe; Gao, Xingming; Yang, Wenqian; Fang, Hao;

Yan, Jun

PATENT ASSIGNEE(S): USA

SOURCE: PCT Int. Appl., 80 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT :	NO.			KIND DATE				APPL	ICAT	ION 1	DATE							
	WO 2005024416				A1	A1 20050317			,	WO 2	004-	 US28	 838	20040907 <					
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		GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	KΖ,	LC,		
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		NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,		
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	RW:	BW,	GH,	GM,	KΕ,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,		
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		SN,	TD,	ΤG															
	US 20070274922						2007	1129		US 2	007-	5708	07	20070531 <					
PRIOR	PRIORITY APPLN. INFO.:								,	US 2	003-	5007	85P		P 20030905 <				
						,	WO 2	004-	US28	838	W 20040907								

OTHER SOURCE(S): MARPAT 142:308776

AB Described herein are boronic acid **fluorescent** compds. and methods of use thereof.

IC ICM G01N033-00

CC 80-3 (Organic Analytical Chemistry)
Section cross-reference(s): 9, 33, 64

ST water soluble boronic acid fluorescent indicator

IT Bacterium (genus)

Biosensors

Blood analysis

Cell

```
Fluorescent indicators
     Formation constant
     Lithiation
    Microtiter plates
    Parasite
     Pharmaceutical analysis
     Tumor markers
     Virus
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
    Antibodies and Immunoglobulins
ΙT
     Blood-group substances
     Carbohydrates, analysis
     DNA
     Glycolipids
     Haptens
     Ligands
     Oligosaccharides, analysis
     RL: ANT (Analyte); ANST (Analytical study)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
    Nucleic acids
     Oligonucleotides
     Peptides, analysis
     RL: ANT (Analyte); ARG (Analytical reagent use); ANST (Analytical study);
     USES (Uses)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
     Dendritic polymers
     Glycoproteins
     Lipids, uses
     Lipopolysaccharides
     Macromolecular compounds
     Polymers, uses
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
    Acids, uses
ΙT
     Group IIIA element compounds
     RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses)
        (boronic acids; analyte detection by fluorometry with water soluble
        boronic acid fluorescent reporter compds.)
TΤ
     RL: ANT (Analyte); ANST (Analytical study)
        (membrane; analyte detection by fluorometry with water soluble boronic
        acid fluorescent reporter compds.)
     50-70-4, Sorbitol, analysis 50-99-7, D-Glucose, analysis 57-48-7,
ΙT
     D-Fructose, analysis 59-23-4, D-Galactose, analysis 63-42-3, Lactose
     147-81-9, Arabinose 2438-80-4, L-Fucose 3458-28-4, D-Mannose
     17598-81-1, Tagatose 71208-06-5, Lewis X
                                                 82993-43-9 92448-22-1,
     Sialyl Lewis a 98603-84-0, Sialyl Lewis X
     RL: ANT (Analyte); ANST (Analytical study)
        (analyte detection by fluorometry with water soluble boronic acid
        fluorescent reporter compds.)
     86-58-8, 8-Quinolineboronic acid 98437-23-1, 2-Benzothienylboronic acid
TΤ
     355386-94-6, 5-Quinolinylboronic acid 371764-64-6,
     4-Quinolinylboronic acid 373384-17-9,
     2-(4-Boronophenyl)-4-quinolinecarboxylic acid
                                                     590417-32-6,
     [6-(Dimethylamino)-2-naphthalenyl]-boronic acid
```

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

ΙT 636987-06-9P, 4-(Dimethylamino)naphthaleneboronic acid 847862-01-5P RL: ARG (Analytical reagent use); PRP (Properties); SPN (Synthetic preparation); ANST (Analytical study); PREP (Preparation); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

4766-33-0P, 5-Amino-1-bromonaphthalene 5328-76-7P, ΙT 5-Nitro-1-bromonaphthalene 10586-45-5P, 5-(Dimethylamino)-1-bromonaphthalene 61047-43-6P, 8-Bromo-2-methylquinoline 847861-96-5P 847861-97-6P

847861-98-7P 847861-99-8P 847862-00-4P

RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

86-57-7, 1-Nitronaphthalene 121-43-7, Trimethylborate 615-36-1, ΤТ 2-Bromoaniline 4170-30-3, Crotonaldehyde 24424-99-5 59557-93-6, 1-Bromo-4-(dimethylamino)naphthalene 201733-56-4 RL: RCT (Reactant); RACT (Reactant or reagent)

(analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

355386-94-6, 5-Quinolinylboronic acid ΙT

RL: ARG (Analytical reagent use); ANST (Analytical study); USES (Uses) (analyte detection by fluorometry with water soluble boronic acid fluorescent reporter compds.)

355386-94-6 ZCAPLUS RN

Boronic acid, B-5-quinolinyl- (CA INDEX NAME) CN



REFERENCE COUNT: 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 45 OF 49 ZCAPLUS COPYRIGHT 2009 ACS on STN 2003:951023 ZCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 140:16738

Preparation of 8-fluoro-3-phenylimidazo[1,2-a]pyridine TITLE:

derivatives as ligands for gamma-aminobutyric acid

(GABA) receptors

INVENTOR(S): Goodacre, Simon Charles; Hallett, David James;

> Humphries, Alexander Charles; Jones, Philip; Kelly, Sarah M.; Merchant, Kevin John; Moore, Kevin William;

Reader, Michael

PATENT ASSIGNEE(S): Merck Sharp & Dohme Limited, UK

SOURCE: PCT Int. Appl., 116 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

										APPLICATION NO.										
														20030523						
		W:	ΑE,	AG,	AL,	AM,	ΑT,	ΑU,	ΑZ,	ВA,	BB,	BG,	BR,	BY,	BZ,	CA,	CH,	CN,		
			CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,		
			GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KR,	KΖ,	LC,	LK,	LR,	LS,		
			LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NI,	NO,	NΖ,	OM,	PH,		
			PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	ТJ,	TM,	TN,	TR,	TT,	TZ,		
			UA,	UG,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW								
		RW:	GH,	GM,	ΚE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,	ΑZ,	BY,		
			KG,	KΖ,	MD,	RU,	ΤJ,	TM,	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,		
			FΙ,	FR,	GB,	GR,	HU,	ΙE,	ΙΤ,	LU,	MC,	NL,	PT,	RO,	SE,	SI,	SK,	TR,		
			BF,	ВJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,	ΤG		
		2486														20030523				
														20030523						
		1511									EP 2	2003-	7276	92		2	0030	523		
	ΕP	1511																		
		R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT,	LI,	LU,	NL,	SE,	MC,	PT,		
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																20030523				
		3230	89			Τ		2006	0415		AT 2	2003-	7276	92	20030523					
		2260																		
	US 20050165048										US 2	2004-	5129	84		2	0041	027		
	US 7279580							2007	1009											
PRIOR	PRIORITY APPLN. INFO.:											2002-								
											WO 2	2003-	GB22	36		W 2	0030	523		
	THER SOURCE(S):					MAR:	PAT	140:	16738	3										
GI	GI																			

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The title compds. [I; X1 represents hydrogen, halogen, C1-6 alkyl, trifluoromethyl, or C1-6 alkoxy; X2 represents hydrogen or halogen; Y represents a chemical bond, an oxygen atom, or a NH or OCH2 linkage; Z represents an optionally substituted aryl or heteroaryl group, or a pyrrolidinonyl group; R1 represents hydrogen, hydrocarbon, a heterocyclic group, halogen, cyano, trifluoromethyl, nitro, ORa, SRa, SORa, SO2Ra, SO2NRaRb, NRaCORb, NRaCO2Rb, CORa, CO2Ra, CONRaRb, or CRa-NORb; and Ra and Rb independently represent hydrogen, hydrocarbon or a heterocyclic group] are prepared These compds. are selective ligands for GABAA receptors, in particular having high affinity for the $\alpha 2$ and/or $\alpha 3$ and/or $\alpha 5$ subunit thereof

and accordingly of benefit in the treatment and/or prevention of neurol. disorders and adverse conditions of the central nervous system, including anxiety, convulsions and cognitive disorders. They were found to possess a Ki value for displacement of [3H]-flumazenil from the $\alpha 2$ and/or $\alpha 3$ and/or $\alpha 5$ subunit of the human GABAA receptor of 100 nM or less.

IT 355386-94-6

RL: RCT (Reactant); RACT (Reactant or reagent) (reactant; preparation of fluorophenylimidazo[a]pyridine derivs. as ligands for GABA receptors)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)



REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 46 OF 49 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2003:178427 ZCAPLUS Full-text

DOCUMENT NUMBER: 138:338121

TITLE: New Synthesis of

1,3-Dihydro-1,4-benzodiazepin-2(2H)-ones and 3-Amino-1,3-dihydro-1,4-benzodiazepin-2(2H)-ones: Pd-Catalyzed Cross-Coupling of Imidoyl Chlorides with

Organoboronic Acids

AUTHOR(S): Nadin, Alan; Sanchez Lopez, Jose M.; Owens, Andrew P.;

Howells, Dean M.; Talbot, Adam C.; Harrison, Timothy Neuroscience Research Centre, Department of Medicinal

CORPORATE SOURCE: Neuroscience Research Centre, Department of Medicinal

Chemistry, Merck, Sharp Dohme Research Laboratories,

Harlow, Essex, CM20 2QR, UK

SOURCE: Journal of Organic Chemistry (2003), 68(7), 2844-2852

CODEN: JOCEAH; ISSN: 0022-3263

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 138:338121

AB A wide variety of functionalized 1,4-benzodiazepines and 3-amino-1,4-benzodiazepines was synthesized via the Pd-catalyzed cross-coupling reaction of an imidoyl chloride with an organometallic reagent as the key step.

IT 355386-94-6

RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation of functionalized dihydrobenzodiazepinones via Pd-catalyzed cross-coupling of imidoyl chlorides with organoboronic acids, amines, or organometallic reagents)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)



REFERENCE COUNT: 50 THERE ARE 50 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 47 OF 49 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2002:123617 ZCAPLUS <u>Full-text</u>

DOCUMENT NUMBER: 136:183819

TITLE: Preparation of (imidazolylalkyl)biphenylcarbonitriles

and analogs as farnesyltransferase inhibitors

INVENTOR(S): Wang, Wei-Bo; Curtin, Michael L.; Fakhoury, Stephen

A.; Gwaltney, Stephen L.; Hasvold, Lisa A.; Hutchins, Charles W.; Li, Qun; Lin, Nan-Horng; Nelson, Lissa Taka Jennings; O'Connor, Steve; Sham, Hing L.;

Sullivan, Gerard M.; Wang, Gary T.; Wang, Xilu

PATENT ASSIGNEE(S): USA

SOURCE: U.S. Pat. Appl. Publ., 189 pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 20020019527	A1	20020214	US 2001-842391	20010425
PRIORITY APPLN. INFO.:			US 2000-200165P P	20000427
OTHER SOURCE(S):	MARPAT	136:183819		

GΙ

AB Title compds. (I) were prepared Thus, 2-MeC6H4C6H3(CN)(CHO)-2,5 was condensed with 1-methyl-2-triethylsilyl-1H-imidazole (preparation each given) and the product 0-arylated to give title compound II. Data for biol. activity of I were given.

IT 355386-94-6P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation of (imidazolylalkyl)biphenylcarbonitriles and analogs as farnesyltransferase inhibitors)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)



L14 ANSWER 48 OF 49 ZCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:798200 ZCAPLUS Full-text

DOCUMENT NUMBER: 135:344482

TITLE: Preparation of substituted

4-(heteroarylmethyl)benzonitriles as

farnesyltransferase inhibitors

INVENTOR(S): Wang, Wei-Bo; Curtin, Michael L.; Fakhoury, Stephen

A.; Gwaltney, Stephen L., II; Hasvold, Lisa A.; Hutchins, Charles W.; Li, Qui; Lin, Nan-Horng; Jennings Nelson, Lissa Taka; O'Connor, Stephen J.; Sham, Hing L.; Sullivan, Gerald M.; Wang, Gary T.;

Wang, Xilu

Abbott Laboratories, USA PATENT ASSIGNEE(S):

SOURCE: PCT Int. Appl., 305 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA'	TENT :	NO.			KIND DATE				APP	LICAT	CION	DATE					
				20011101 20020523			WO	2001-		20010425							
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		LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX	, MZ,	NO,	NZ,	PL,	PT,	RO,	RU,
		SD,	SE,	SG,	SI,	SK,	SL,	ΤJ,	TM,	TR	TT,	TZ,	UA,	UG,	UZ,	VN,	YU,
	ZA, ZW																
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		ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GW,	ML	, MR,	NE,	SN,	TD,	ΤG		
CA	2407	093			A1 20011101					CA	2001-	-2407	20010425				
EP	1276	726			A2 20030122				EΡ	2001-	-9327	20010425					
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		ΙE,	SI,	LT,	LV,	FI,	RO,	MK,	CY,	AL	, TR						
JP	2004	5090	64		Τ		2004	0325		JΡ	2001-	-5784	20010425				
MX	2002	0106	08		Α		2003	0514		MX	2002-	-1060	8		2	0021	025
PRIORIT	IORITY APPLN. INFO.:									US	2000-	-5632	56		A 2	0000	427
										US	2001-	-8222	05		A 2	0010	402
										WO	2001-	-US13	678	1	W 2	0010	425
OTHER SO	• •					PAT	135:	3444	82								

GΙ

AB The title compds. [I; A1 = (un)substituted alkylene, etc.; R1 = halo, cycloalkyl, aryl, heteroaryl; R2 = heteroaryl selected from imidazolyl, pyrazolyl, pyrrolyl, etc.] and their pharmaceutically acceptable salts which farnesyltransferase, were prepared E.g., 3-step synthesis of the benzonitrile II.HCl which 88% inhibition of farnesyltransferase at 10-6 M, was given.

IT 355386-94-6P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation of substituted 4-(heteroarylmethyl)benzonitriles as farnesyltransferase inhibitors)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)



REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L14 ANSWER 49 OF 49 ZCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2001:618000 ZCAPLUS Full-text

DOCUMENT NUMBER: 135:180708

TITLE: Preparation and formulation of biarylcarboxamides as

nicotinic acetylcholine receptor agonists for therapeutic use in the treatment or prophylaxis of psychotic and intellectual impairment disorders

INVENTOR(S): Phillips, Eifion; Schmiesing, Richard

PATENT ASSIGNEE(S): Astrazeneca Ab, Swed. SOURCE: PCT Int. Appl., 84 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PA.	TENT	NO.			KINE)		APPLICATION NO.												
									WO 2001-SE329											
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OTHER SO	OURCE	(S):			MARE	PAT	135:	18070	8 C											
GI																				

52

$$\mathbb{R} \xrightarrow{\mathbb{C}_{\mathcal{O}}} \mathbb{C}_{\mathcal{O}} \xrightarrow{\mathbb{N}} \mathbb{C}_{\mathbb{N}}$$

Biarylcarboxamides, such as A-NH-C(:X)-Ar1-Ar2 {A = 1-azabicyclo[2.2.2]octan-3-yl, 1-azabicyclo[2.2.1]heptan-3-yl, 7-azabicyclo[2.2.1]heptan-2-yl, 2-azabicyclo[2.2.2]octan-4-yl; Ar1, Ar2 = aryl, heteroaryl, substituted aryl, substituted heteroaryl; X = 0, S}, were prepared for pharmaceutical use as nicotinic acetylcholine receptor agonists for treatment or prophylaxis of psychotic and intellectual impairment disorders. Thus, biarylcarboxamide I (R = Ph) was prepared via aromatic coupling of the hydrochloride salt of the corresponding bromide I (R = Br) with phenylboronic acid using Pd(PPh3)4 and cesium carbonate in a mixt of 1,2-dimethoxyethane, ethanol, and water. The prepared biarylcarboxamides were assayed for binding affinity for the α 7AChR and α 4AChR nicotinic acetylcholine receptors. Pharmaceutical formulations for delivery of the biarylcarboxamides were also discussed.

IT 355386-94-6

RL: RCT (Reactant); RACT (Reactant or reagent)
(preparation and formulation of biarylcarboxamides as nicotinic acetylcholine receptor agonists for therapeutic use in the treatment or prophylaxis of psychotic and intellectual impairment disorders)

RN 355386-94-6 ZCAPLUS

CN Boronic acid, B-5-quinolinyl- (CA INDEX NAME)

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L15

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             STRUCTURE UPLOADED
L1
L2
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    FILE 'STNGUIDE' ENTERED AT 15:19:37 ON 31 MAR 2009
    FILE 'ZCAPLUS' ENTERED AT 15:19:47 ON 31 MAR 2009
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24 DUP REM L51 L50 (17 DUPLICATES REMOVED)

L52

ANSWERS '1-22' FROM FILE ZCAPLUS ANSWERS '23-24' FROM FILE BIOSIS

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D IALL L52 23-24

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FILE 'ZCAPLUS' ENTERED AT 15:48:24 ON 31 MAR 2009

D STAT QUE L12

D IBIB ABS HITSTR L12 1-5

FILE 'REGISTRY' ENTERED AT 15:49:15 ON 31 MAR 2009

FILE 'ZCAPLUS' ENTERED AT 15:49:24 ON 31 MAR 2009

D STAT QUE L29

D STAT QUE L13

D IBIB ABS HITIND HITSTR L29 1-4

D IBIB ABS HITSTR L13 102-106

FILE 'REGISTRY' ENTERED AT 15:50:33 ON 31 MAR 2009

FILE 'ZCAPLUS' ENTERED AT 15:50:36 ON 31 MAR 2009

D STAT QUE L34

D STAT QUE L14

D IBIB ABS HITIND HITSTR L34 1

D IBIB ABS HITSTR L14 45-49

FILE HOME

FILE REGISTRY

Property values tagged with IC are from the ${\tt ZIC/VINITI}$ data file provided by ${\tt InfoChem.}$

STRUCTURE FILE UPDATES: 29 MAR 2009 HIGHEST RN 1129300-01-1 DICTIONARY FILE UPDATES: 29 MAR 2009 HIGHEST RN 1129300-01-1

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FILE ZCAPLUS

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FILE COVERS 1907 - 31 Mar 2009 VOL 150 ISS 14 FILE LAST UPDATED: 30 Mar 2009 (20090330/ED)

ZCAplus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2008.

CAS Information Use Policies apply and are available at:

http://www.cas.org/legal/infopolicy.html

This file contains CAS Registry Numbers for easy and accurate substance identification.

FILE STNGUIDE

FILE CONTAINS CURRENT INFORMATION.

LAST RELOADED: Mar 27, 2009 (20090327/UP).

FILE MEDLINE

FILE LAST UPDATED: 28 Mar 2009 (20090328/UP). FILE COVERS 1949 TO DATE.

MEDLINE and LMEDLINE have been updated with the 2009 Medical Subject Headings (MeSH) vocabulary and tree numbers from the U.S. National Libra of Medicine (NLM). Additional information is available at

http://www.nlm.nih.gov/pubs/techbull/nd08/nd08_medline_data_changes_2009.

On February 21, 2009, MEDLINE was reloaded. See HELP RLOAD for details.

This file contains CAS Registry Numbers for easy and accurate substance identification.

See HELP RANGE before carrying out any RANGE search.

FILE EMBASE

FILE COVERS 1974 TO 31 Mar 2009 (20090331/ED)

EMBASE was reloaded on March 30, 2008.

EMBASE is now updated daily. SDI frequency remains weekly (default) and biweekly.

This file contains CAS Registry Numbers for easy and accurate substance identification.

Beginning January 2008, Elsevier will no longer provide EMTREE codes as part of the EMTREE thesaurus in EMBASE. Please update your current-awareness alerts (SDIs) if they contain EMTREE codes.

For further assistance, please contact your local helpdesk.

FILE BIOSIS

FILE COVERS 1926 TO DATE.

CAS REGISTRY NUMBERS AND CHEMICAL NAMES (CNs) PRESENT FROM JANUARY 1926 TO DATE.

RECORDS LAST ADDED: 25 March 2009 (20090325/ED)

BIOSIS has been augmented with 1.8 million archival records from 1926 through 1968. These records have been re-indexed to match current BIOSIS indexing.

FILE WPIX

FILE LAST UPDATED: 23 MAR 2009 <20090323/UP>
MOST RECENT UPDATE: 200918 <200918/DW>

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>>> IPC and US National Classifications have been updated
 with reclassifications to the end of 2008.
 ECLA, F-Term and FI-Term classifications are complete
 to the end of 2008.
 No update date (UP) has been created for the reclassified

No update date (UP) has been created for the reclassified documents, but they can be identified by specific update codes (see HELP CLA for details) <<<

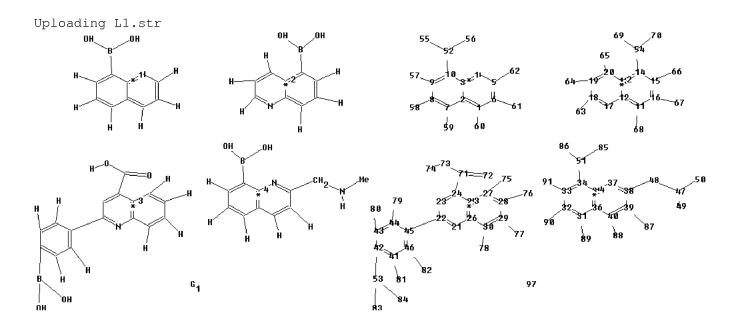
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EXPLORE DERWENT WORLD PATENTS INDEX IN STN ANAVIST, VERSION 2.0: http://www.stn-international.com/DWPIAnaVist2_0608.html

>>> HELP for European Patent Classifications see HELP ECLA, HELP ICO <<<



chain nodes :

ring nodes :

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24 \quad 25 \quad 26 \quad 27 \quad 28 \quad 29 \quad 30 \quad 31 \quad 32 \quad 33 \quad 34 \quad 35 \quad 36 \quad 37 \quad 38 \quad 39 \quad 40 \quad 41 \quad 42 \quad 43 \quad 44
chain bonds :
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20-65 22-45 24-71 27-75 28-76 29-77 30-78 31-89 32-90 33-91 34-51 38-48
39-87 40-88
41-81 42-53 43-80 44-79 46-82 47-48 47-49 47-50 51-85 51-86 52-55 52-56
53-83 53-84
54-69 54-70 71-72 71-73 73-74
ring bonds :
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13 - 14 \quad 13 - 20 \quad 14 - 15 \quad 15 - 16 \quad 17 - 18 \quad 18 - 19 \quad 19 - 20 \quad 21 - 22 \quad 21 - 26 \quad 22 - 23 \quad 23 - 24 \quad 24 - 25
25-26 25-27
26-30 27-28 28-29 29-30 31-32 31-36 32-33 33-34 34-35 35-36 35-37 36-40
37-38 38-39
39-40 41-42 41-46 42-43 43-44 44-45 45-46
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54-69 54-70 73-74
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G1:[*1],[*2],[*3],[*4]

Match level :

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